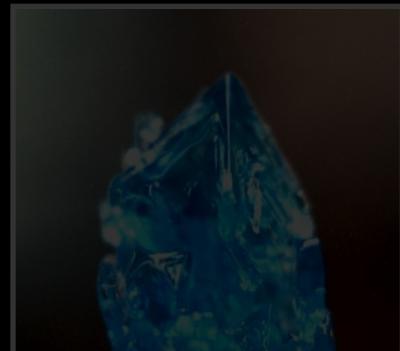


ENERGY GEOTECHNOLOGY

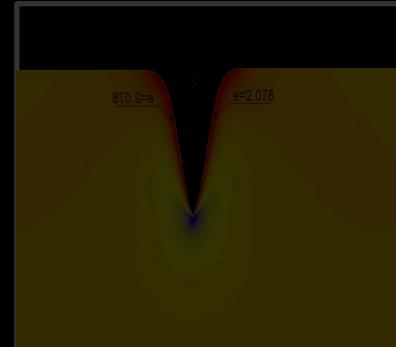
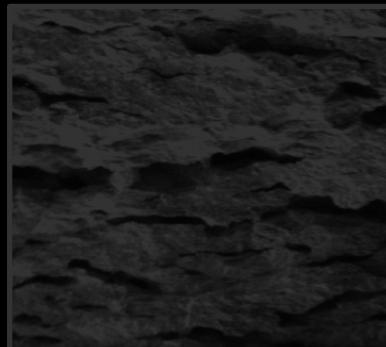
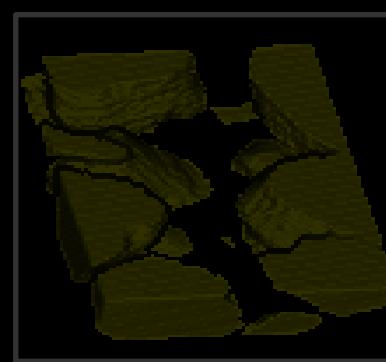
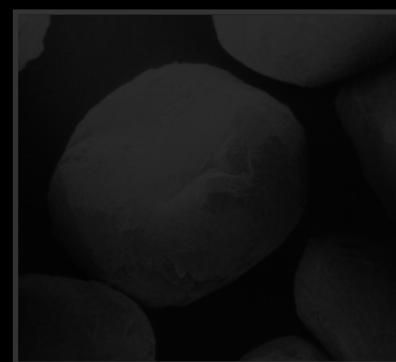
Mixed Fluid Conditions

J. Carlos Santamarina and Jaewon Jang





energy geotechnology

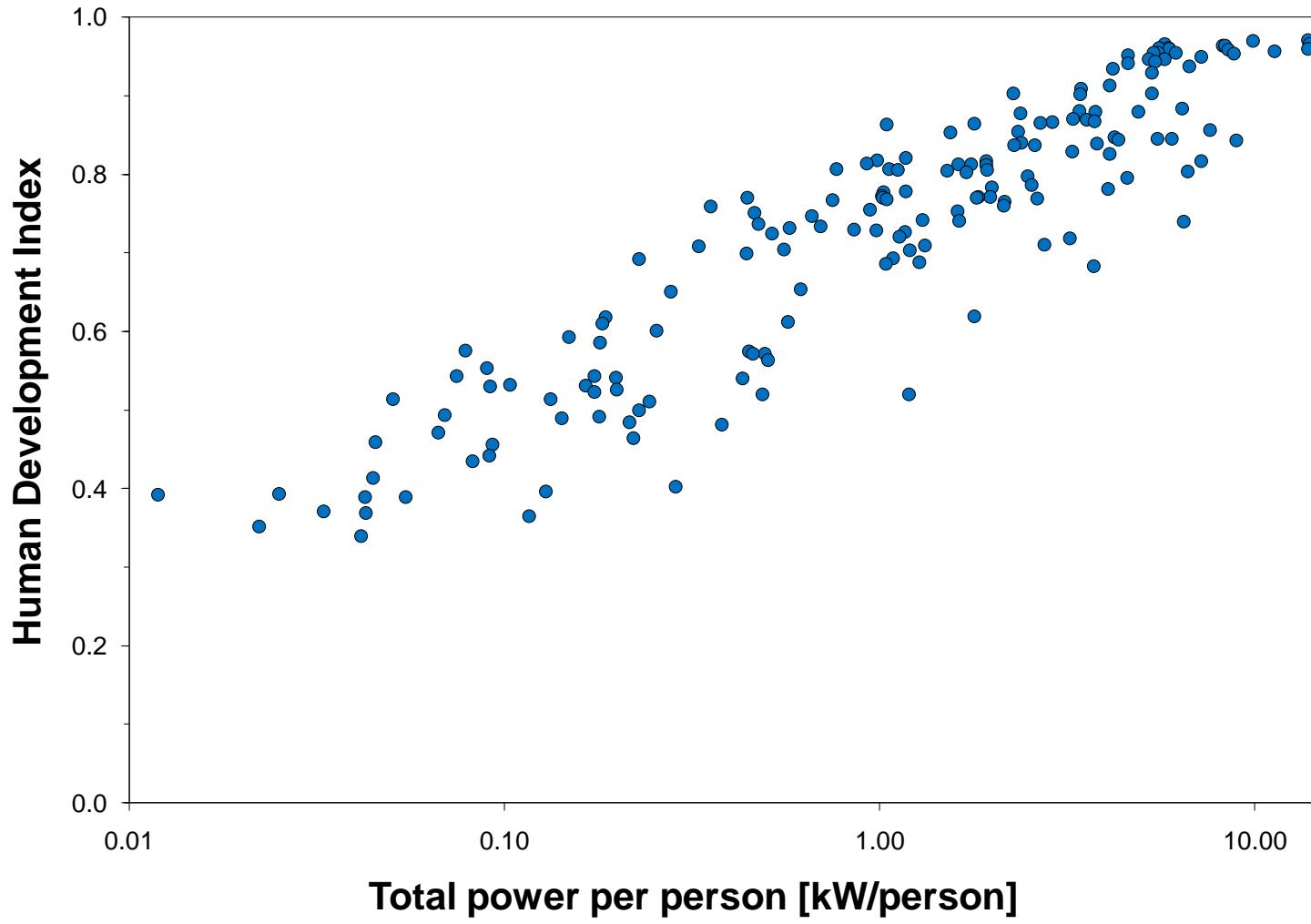


Energy in the News



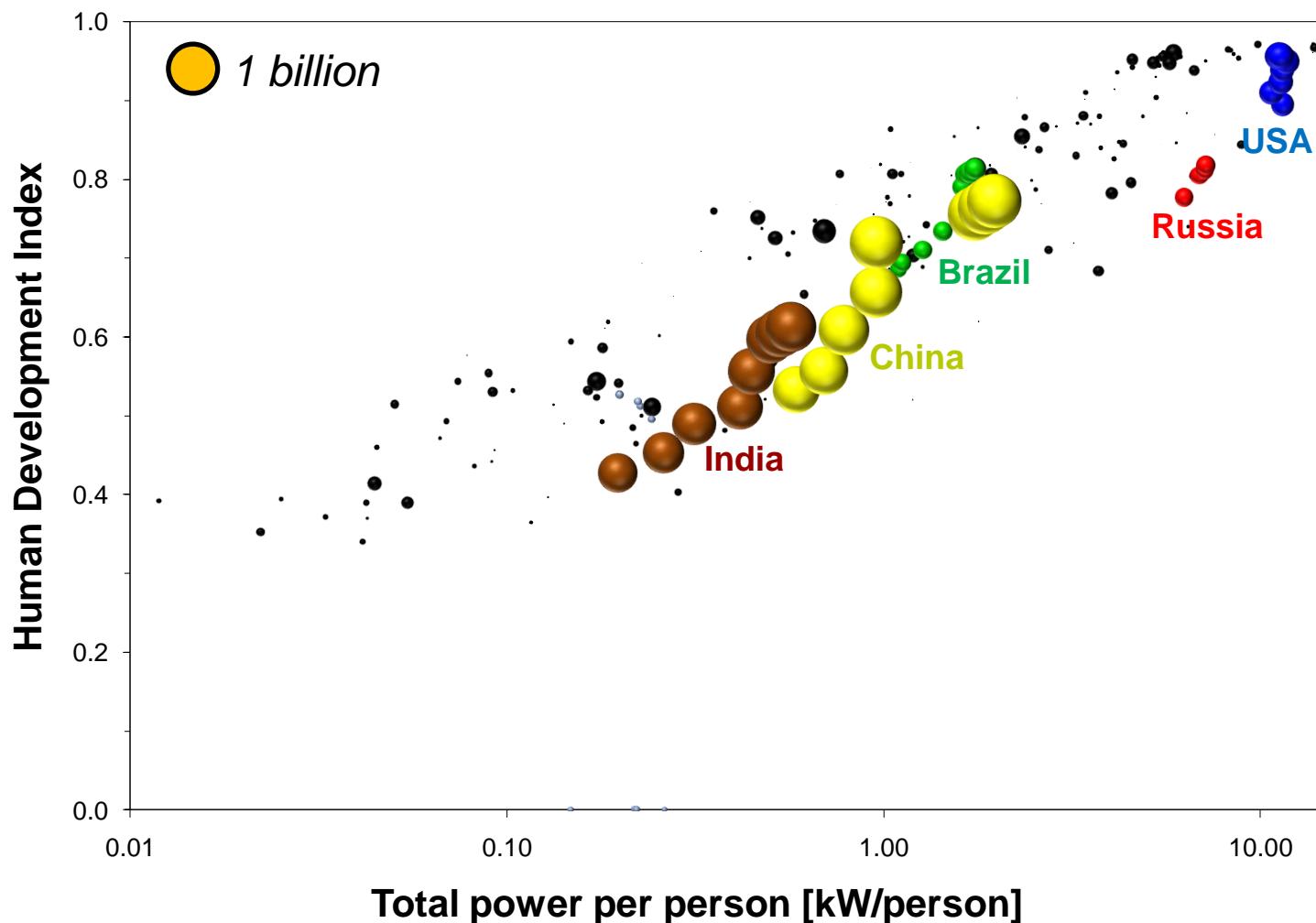
Energy and Life

(Global: 2008)



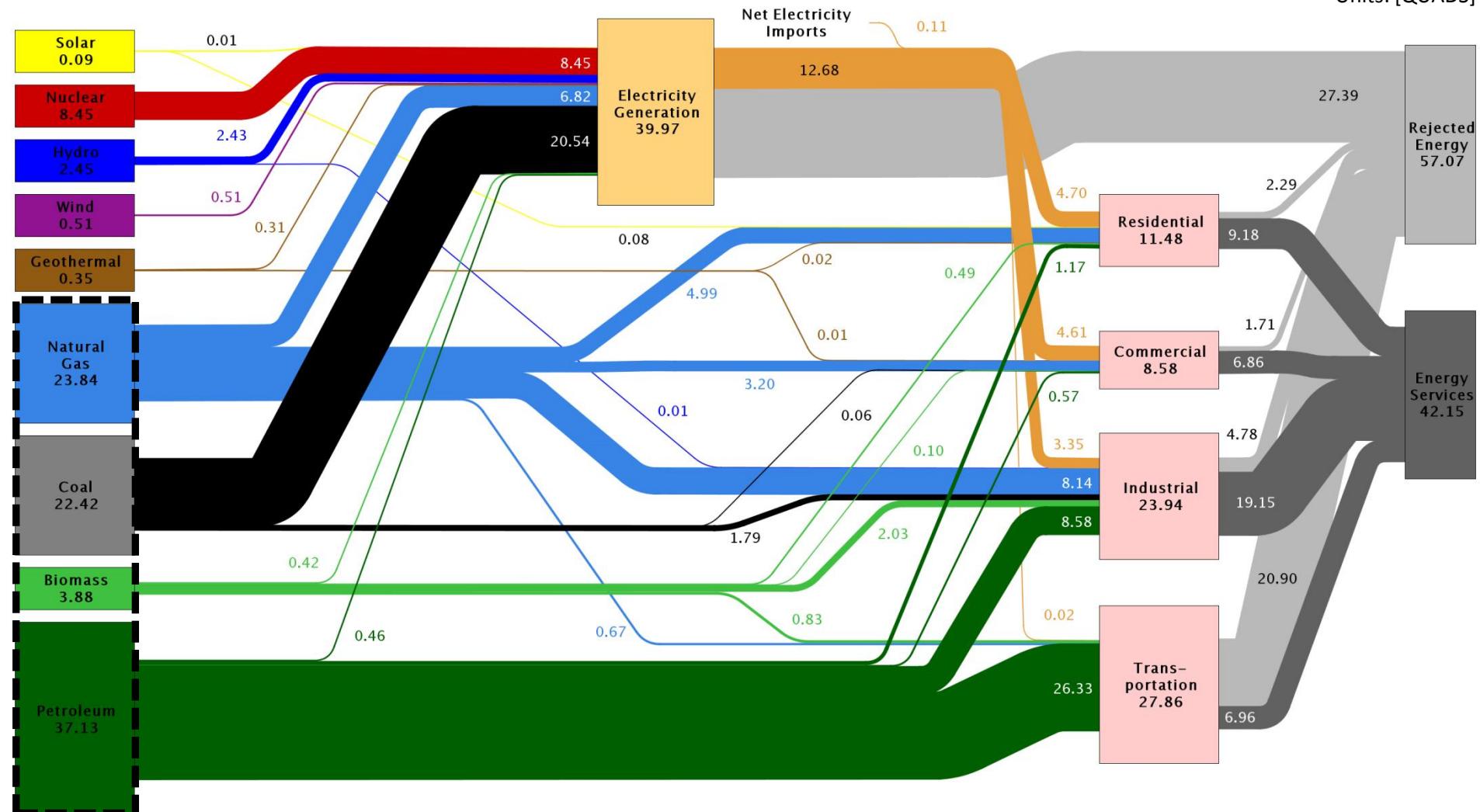
Energy and Life

(Global: 2008 – BRIC trends: 1980-2007)

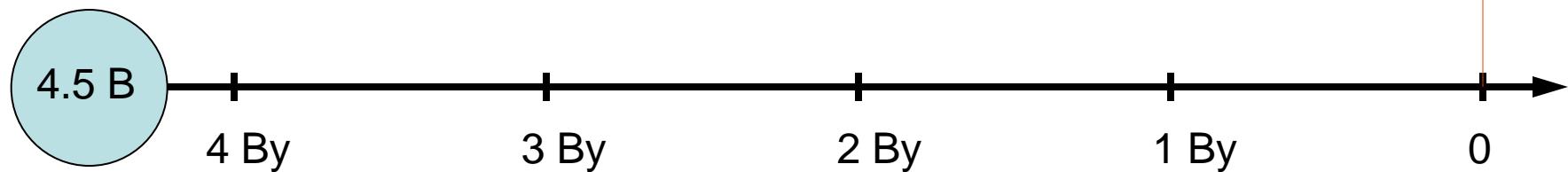


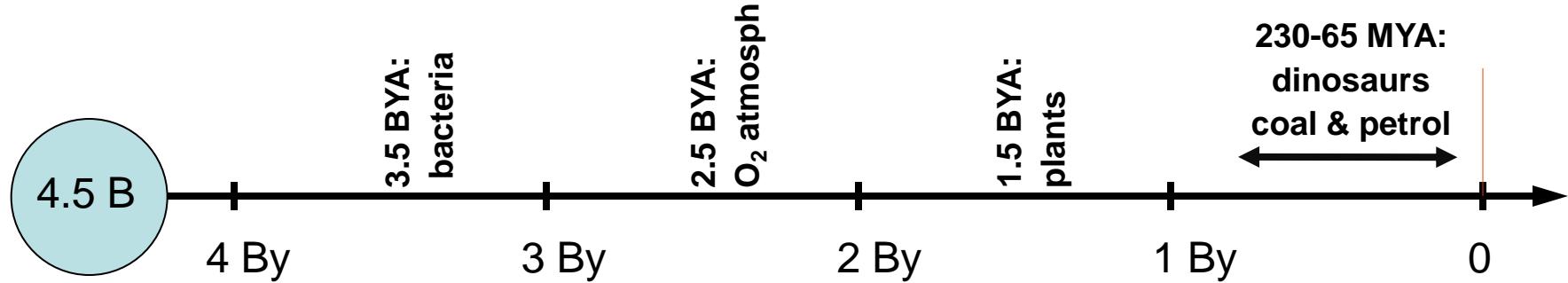
Sources – Case: USA

2008 LLNL – DOE
Units: [QUADS]



Geo-Centered Perspective: Time Scale





4.5 B

4 By

3 By

2 By

1 By

0

10 My

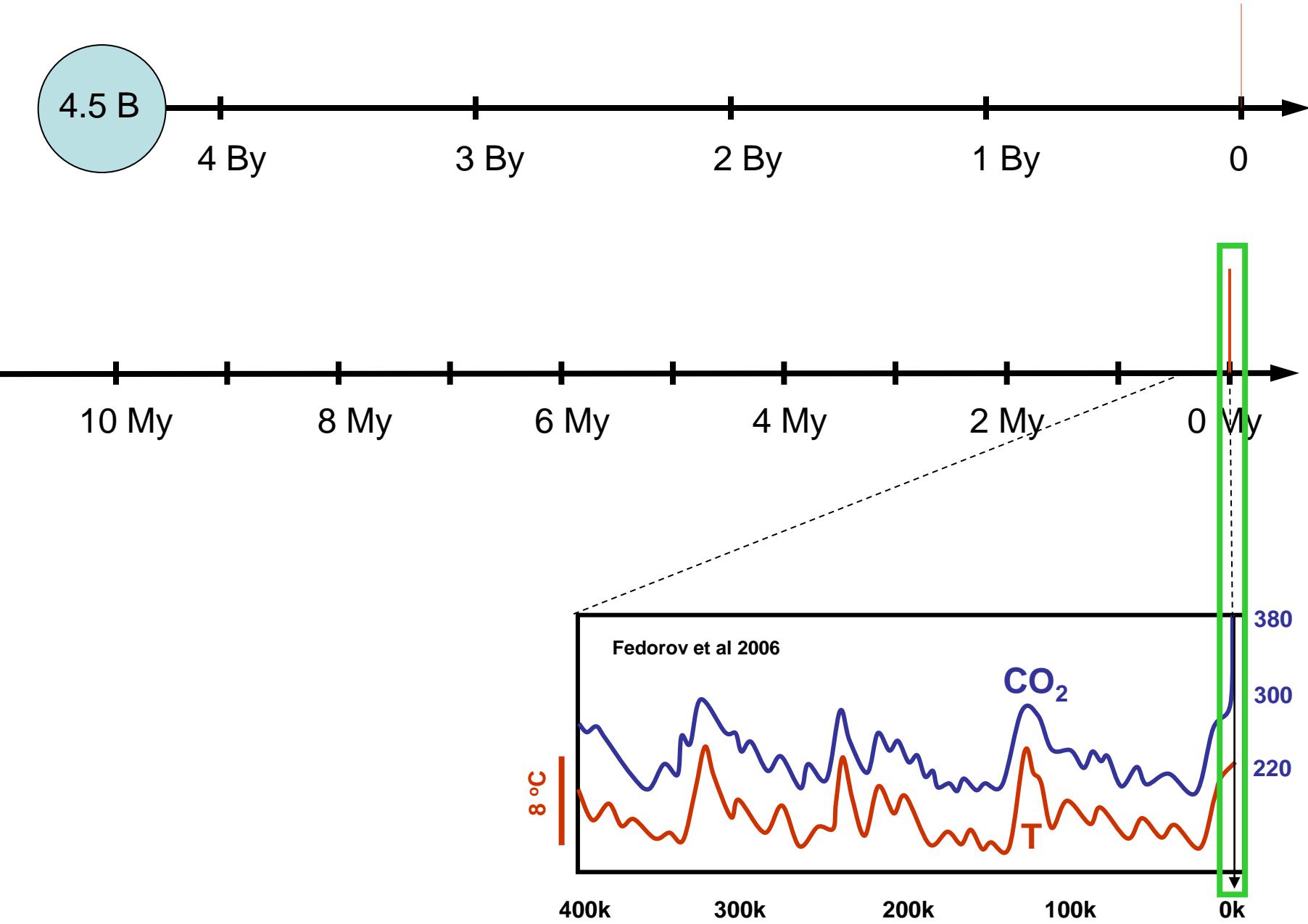
8 My

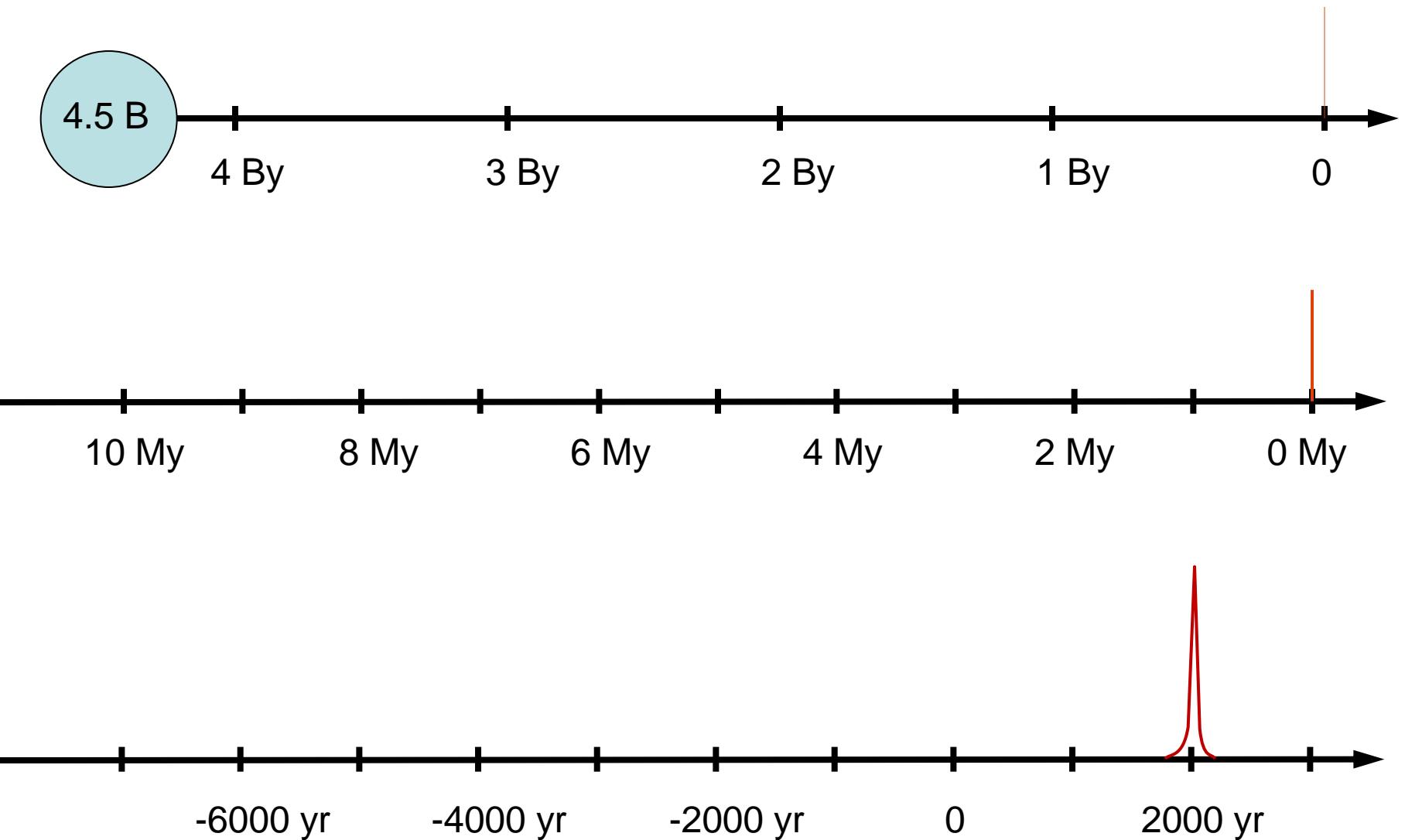
6 My

4 My

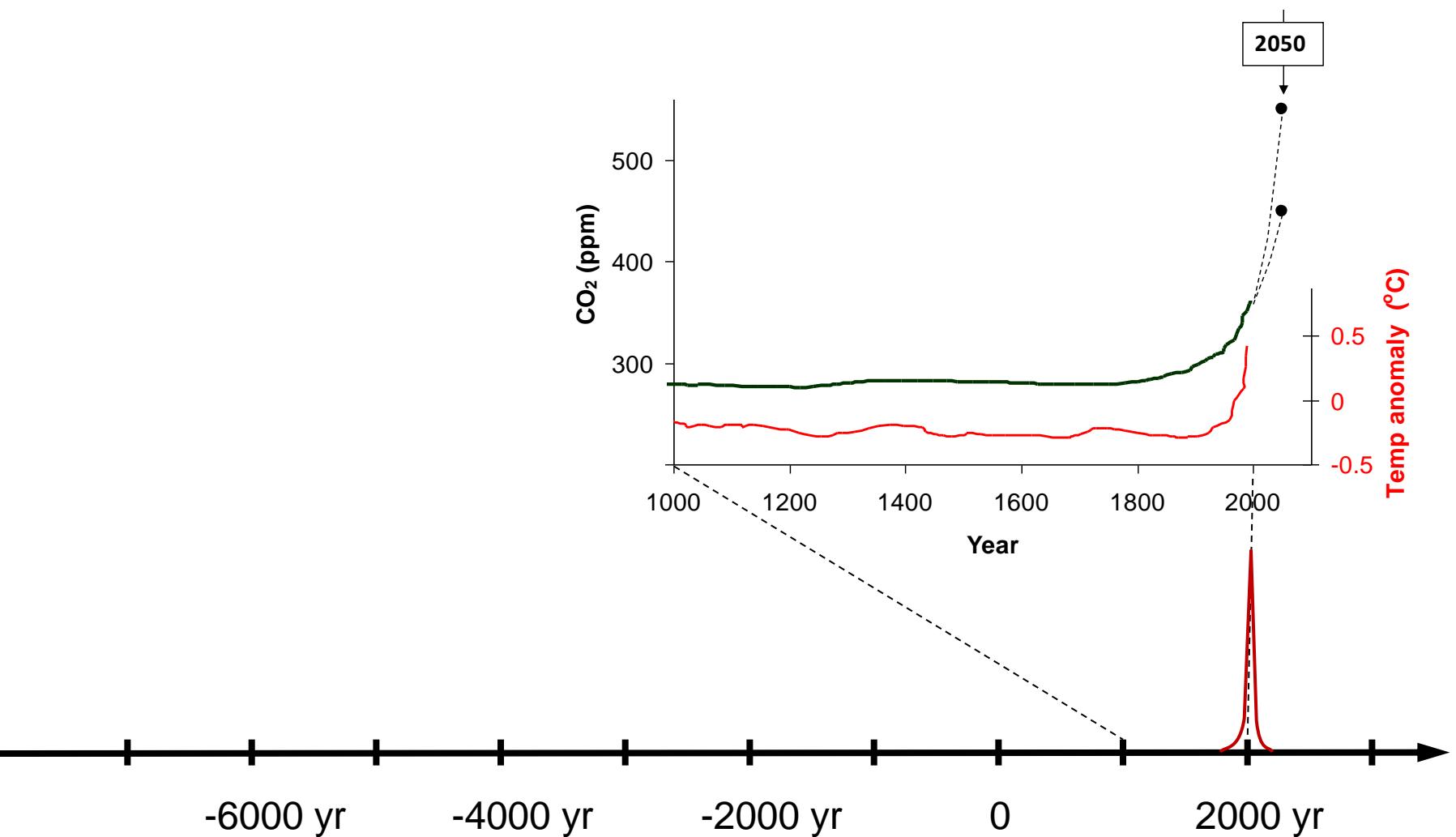
2 My

0 My





History of fossil fuels: a δ -function



Global implications

Geo-Centered Perspective: Spatial Scale



Energy Geotechnology

FOSSIL FUELS (C-BASED)			RENEWABLE			Nuclear				
Petroleum	Gas	Coal	Wind	Solar	Biofuels					
<ul style="list-style-type: none"> • fines & clogging • sand production • shale instability • EOR • heavy oil & tar sand <ul style="list-style-type: none"> • mixed fluid flow, percolation • contact angle & surface tension = $f(u_a)$ 	<ul style="list-style-type: none"> • gas hydrates • gas storage • low-T LNG found. 	<ul style="list-style-type: none"> • characterization • optimal extraction • subsurface conv. 	<ul style="list-style-type: none"> • periodic load • ratcheting 			<ul style="list-style-type: none"> • engineered soils • decommission • leak detect • leak repair 				
GEOLOGICAL STORAGE										
CO₂ sequestration 10 ⁴ -10 ⁵ yr BTHCM mineral dissolution → shear faults, pipes			Energy Storage CAES, phase-change Cyclic HTCM		Waste storage 10 ⁵ yr BTHCM					
GEO-ENVIRONMENTAL REMEDIATION										
CONSERVATION										

- Hydro-electric: capacity almost saturated

Energy Geotechnology: Phases

Gas

water vapor

CO_2

CH_4

- *supercritical CO_2*

Liquid

water

CO_2

oil

Solid

mineral

ice

CO_2 hydrate

CH_4 hydrate

Summary: Energy Geotechnology

Quality of life

Current development patterns: HDI ↔ Energy

Current

15.6 TW – increasing at ~1% per year

Fossil fuels

Stored solar energy (1 billion years in the making)

C-Economy: ~300 years

Short-term: C-emissions → Climate (global)

CO₂ storage

Fossil fuels more sustainable ... but...

Energy

Geotechnology

resource recovery

production

energy storage

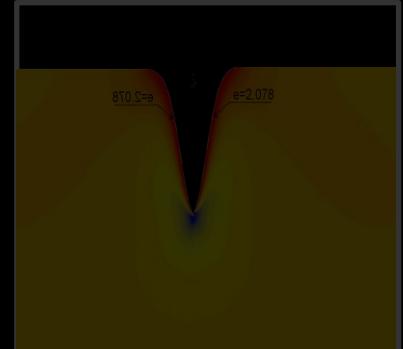
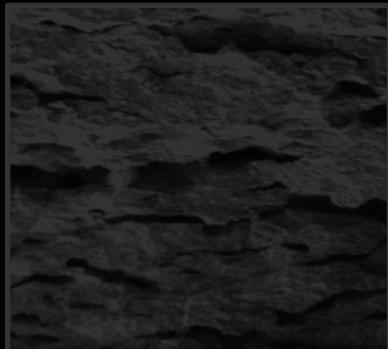
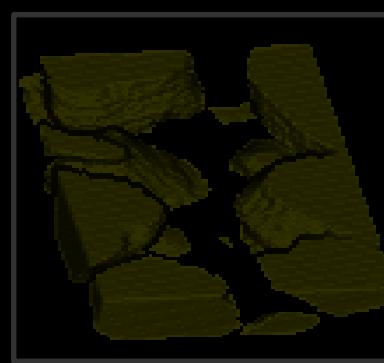
waste storage

efficiency

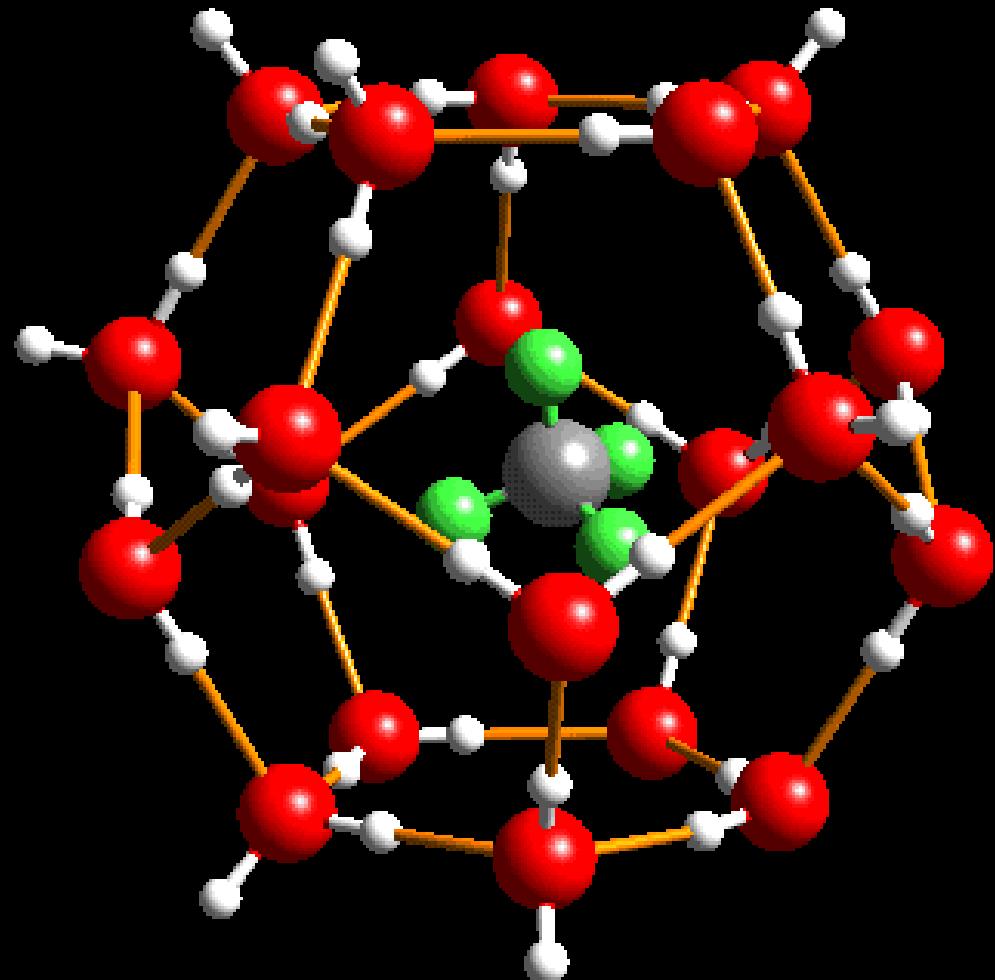
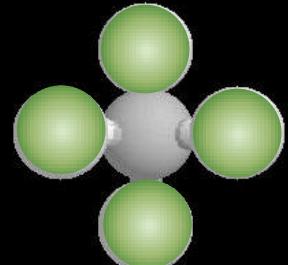
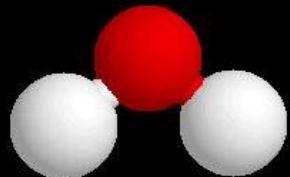
Wide range of multi-phase conditions



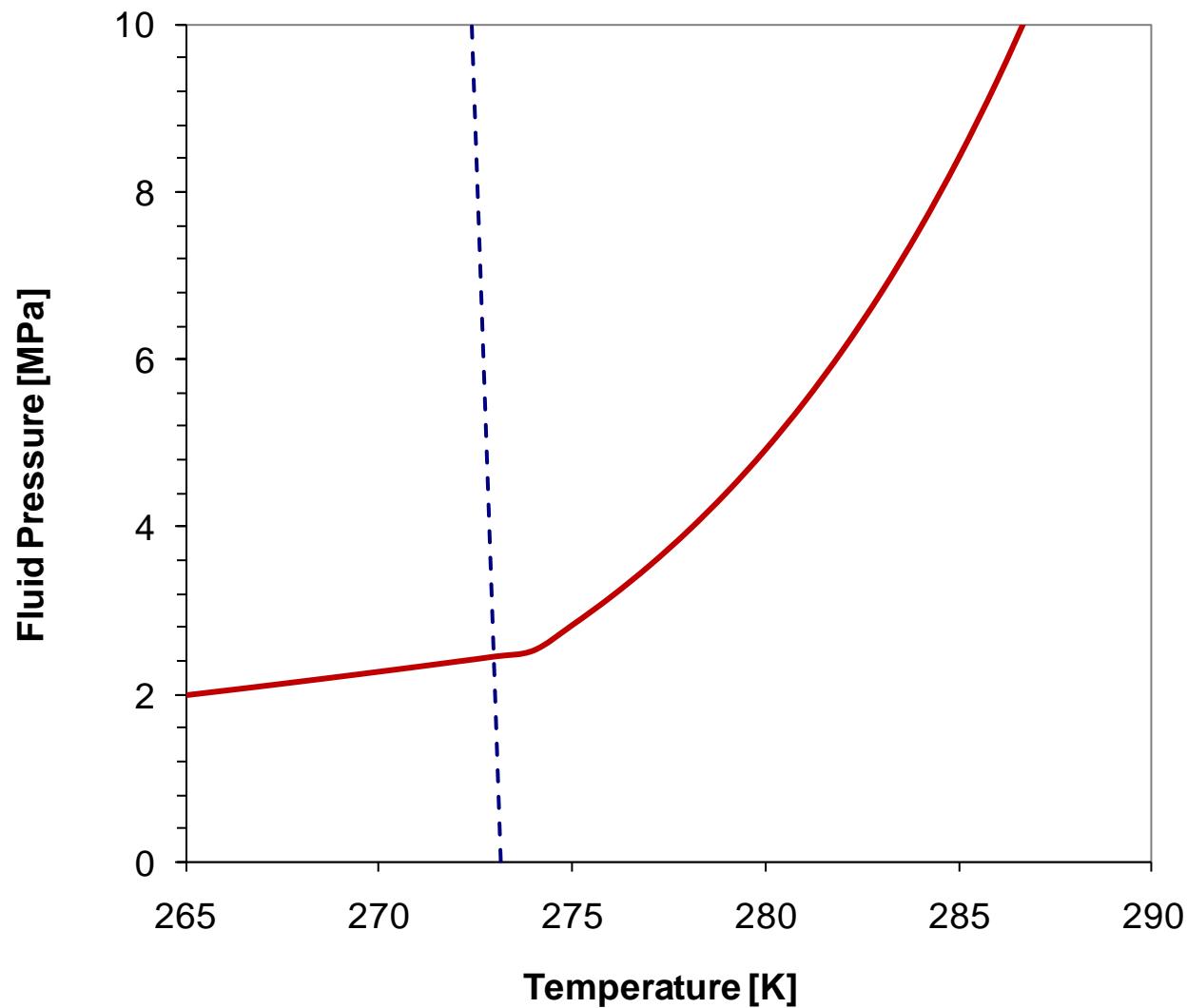
CH₄ hydrates



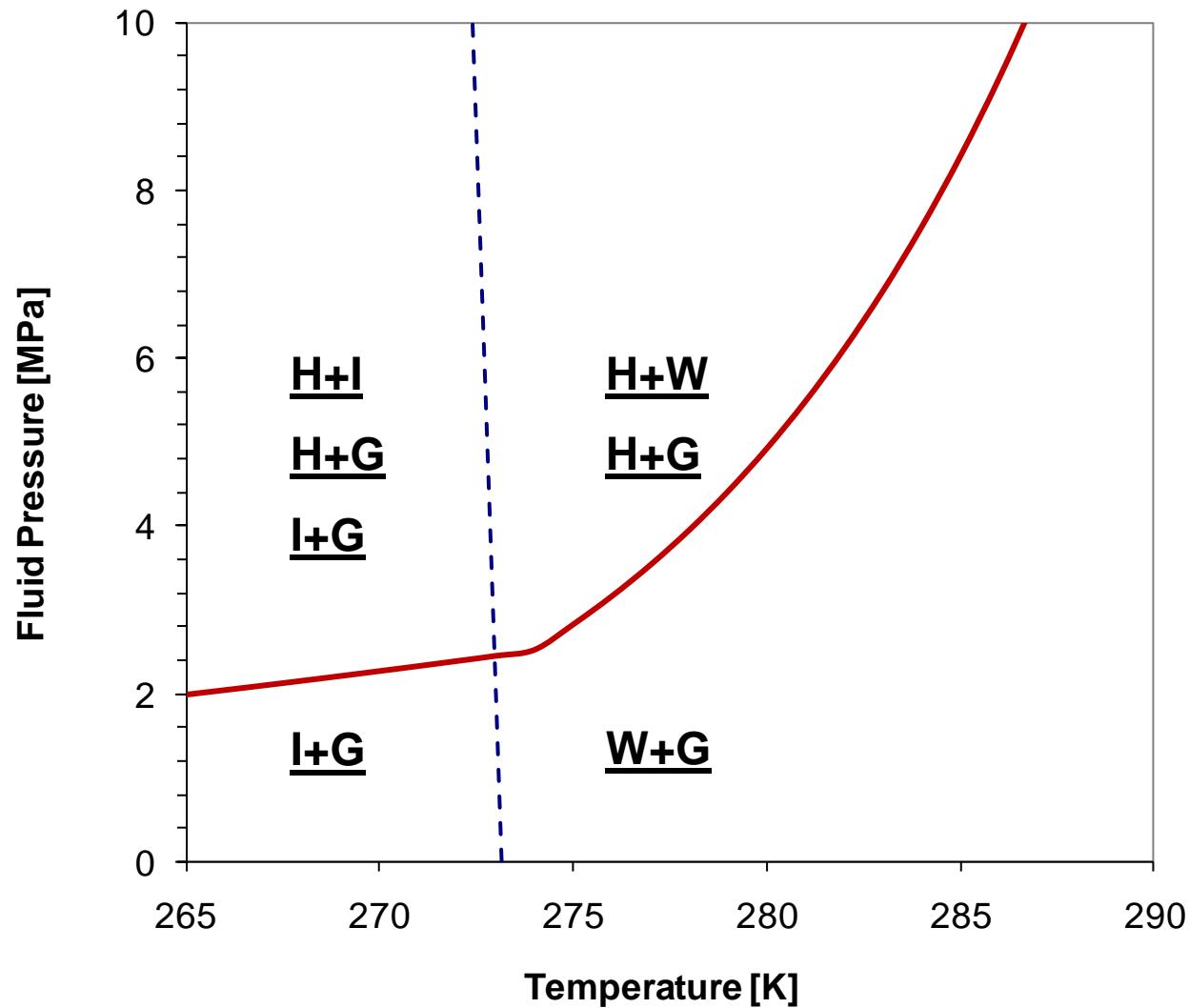
Hydrates (clathrate = cage)



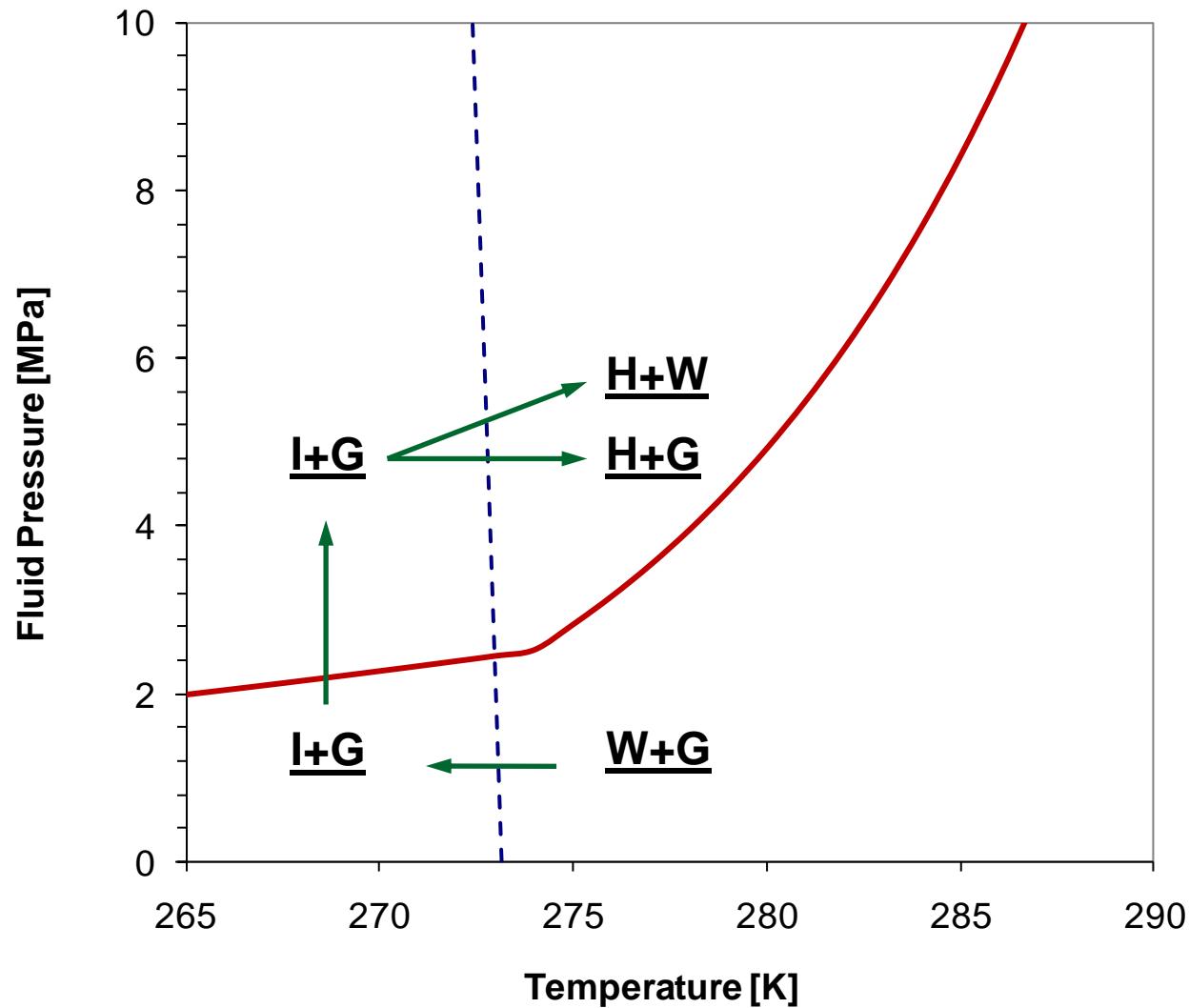
Methane Hydrate



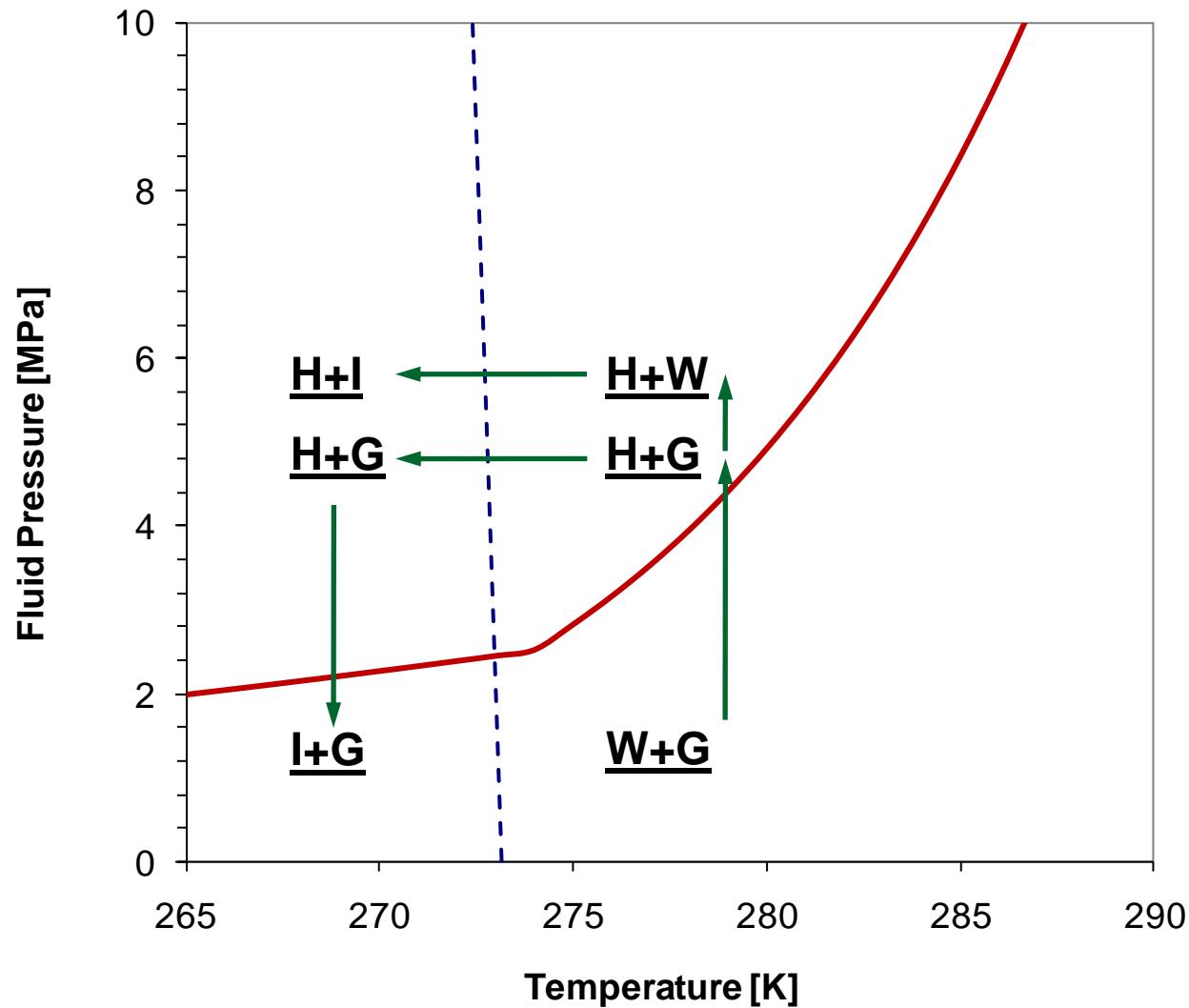
Methane Hydrate



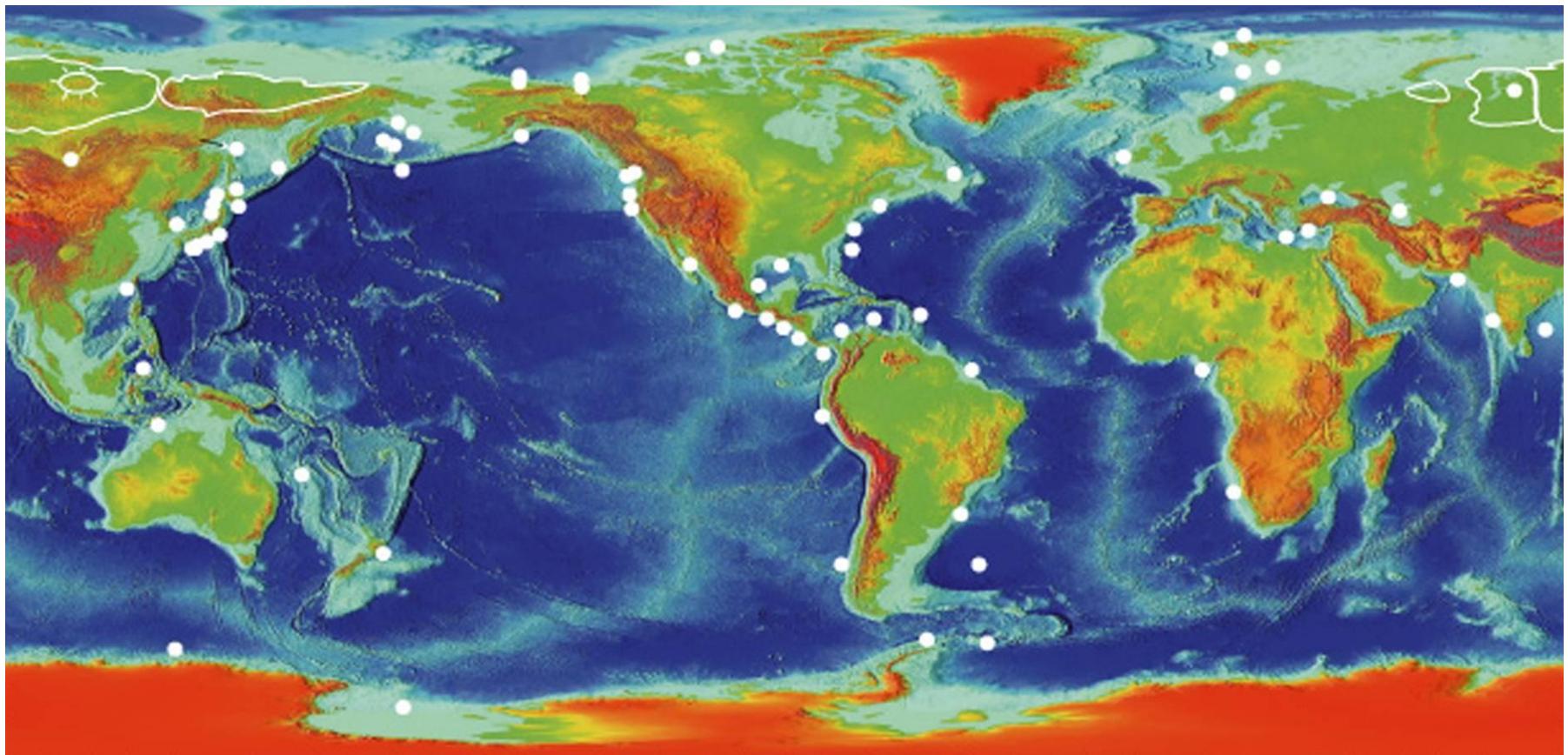
Methane Hydrate



Methane Hydrate



Methane Hydrate - Occurrence



Hydrate – Key Observations

CH_4 solubility in H_2O

$\text{CH}_4 : 750 \text{ H}_2\text{O}$

CH_4 concentration in hydrate

$\text{CH}_4 : 6 \text{ H}_2\text{O}$

Diffusivity CH_4 in

water

$\sim 1 \times 10^{-9} \text{ m}^2/\text{s}$

hydrate

$\sim 5 \times 10^{-13} \text{ m}^2/\text{s}$

Ice \rightarrow water

$V_w/V_{\text{ice}} = 0.92$

Hydrate \rightarrow (water+ CH_4^{gas})

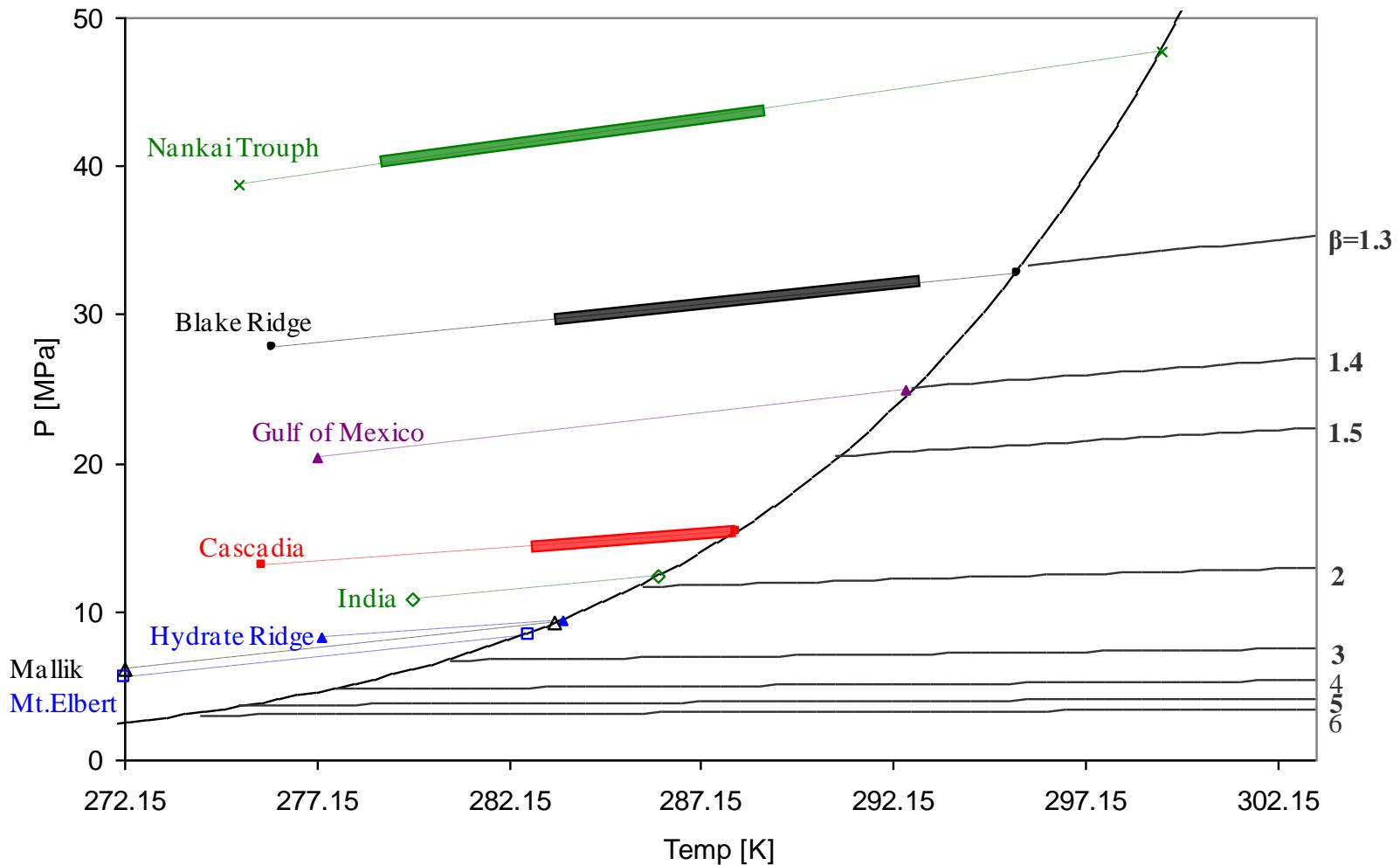
$V_{w+\text{g}}/V_{\text{hyd}} = 1 \text{ to } >6$

*formation:
 CH_4 diffusion*

production

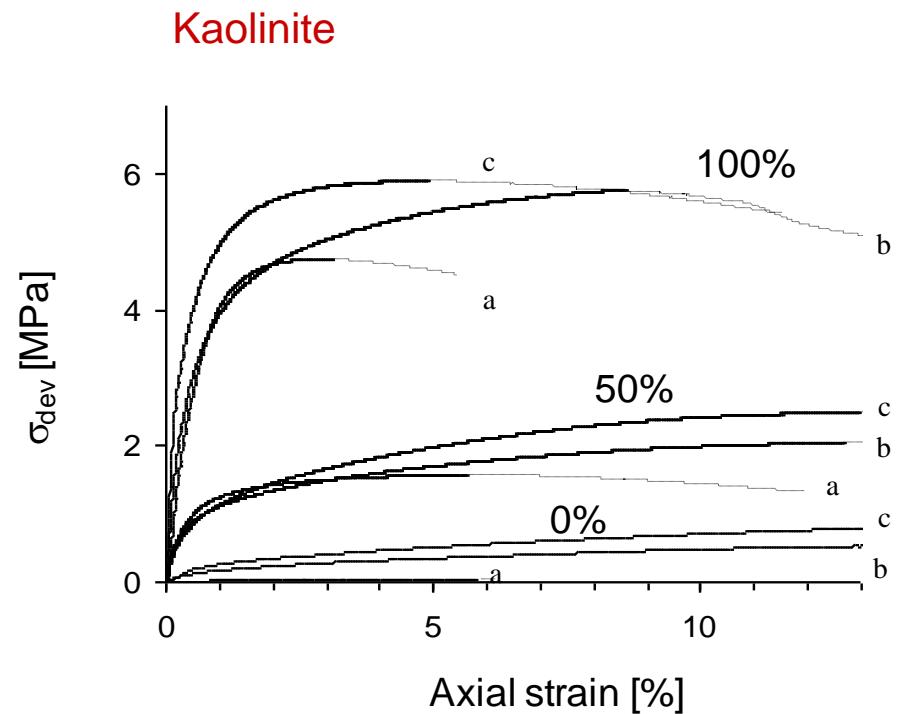
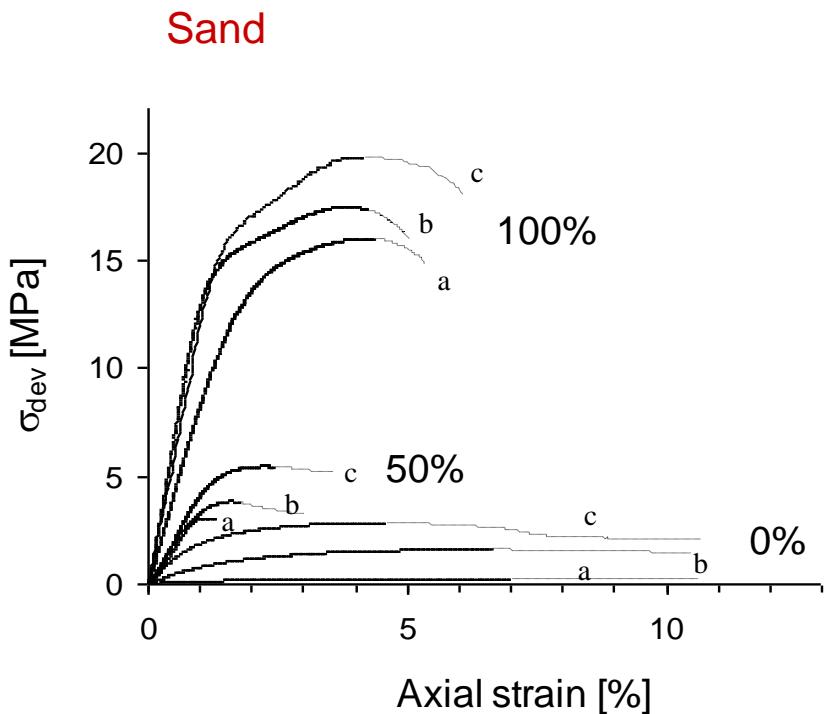
Fluid Volume Expansion

$$\beta = \frac{V_W + V_G}{V_{hyd}}$$



a: $\sigma'_c=0.03$ MPa
b: $\sigma'_c=0.5$ MPa
c: $\sigma'_c=1$ MPa

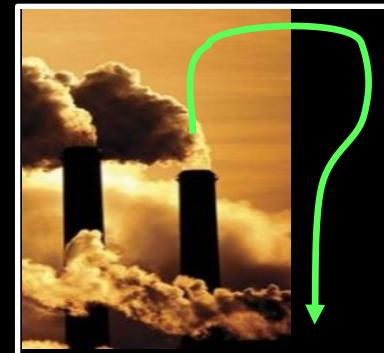
Hydrate-bearing Sediments



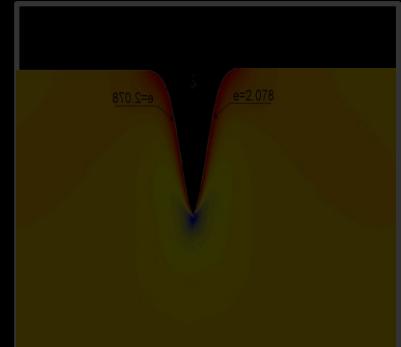
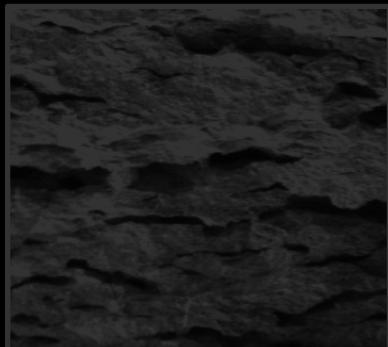
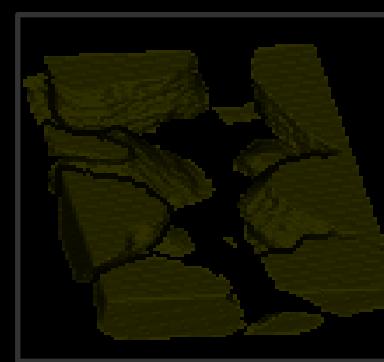
all THCEM properties = non lineal functions of S_{hyd}

Summary: Methane Hydrates

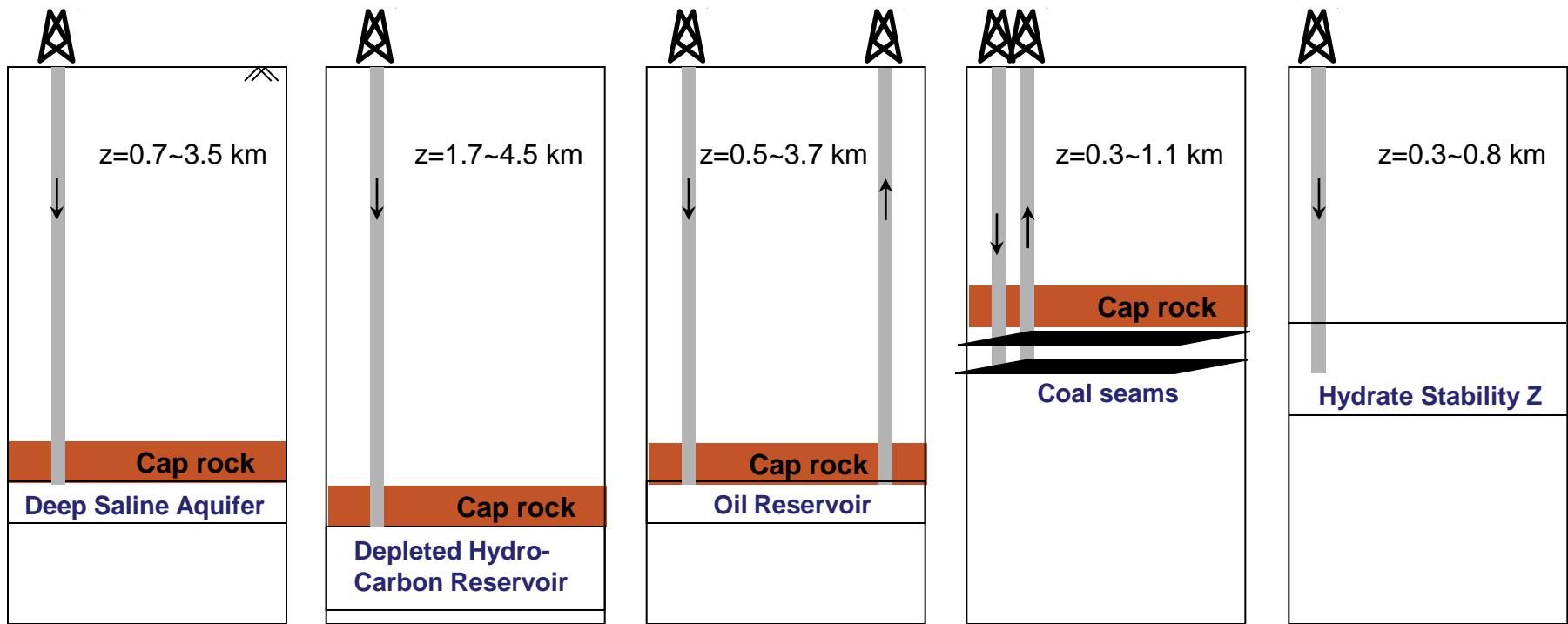
Relevance:	C-reserves	climate change	instability
Formation	PT history dependent S_{hyd} is CH ₄ -limited (typically)		
Multi-phase	<u>Hydrate</u> <u>Water</u> <u>Gas</u> <u>Ice</u> <u>Mineral</u> (not all at once)		
Pore habit	Patchy (coarse grained sediments) Lenses (fine grained sediments)		
THCEM properties	Non linear functions of S_{hyd}		
Gas Production	Endothermic (may be heat-limited) Very large volume expansion Production from sands? from clayey sediments?		



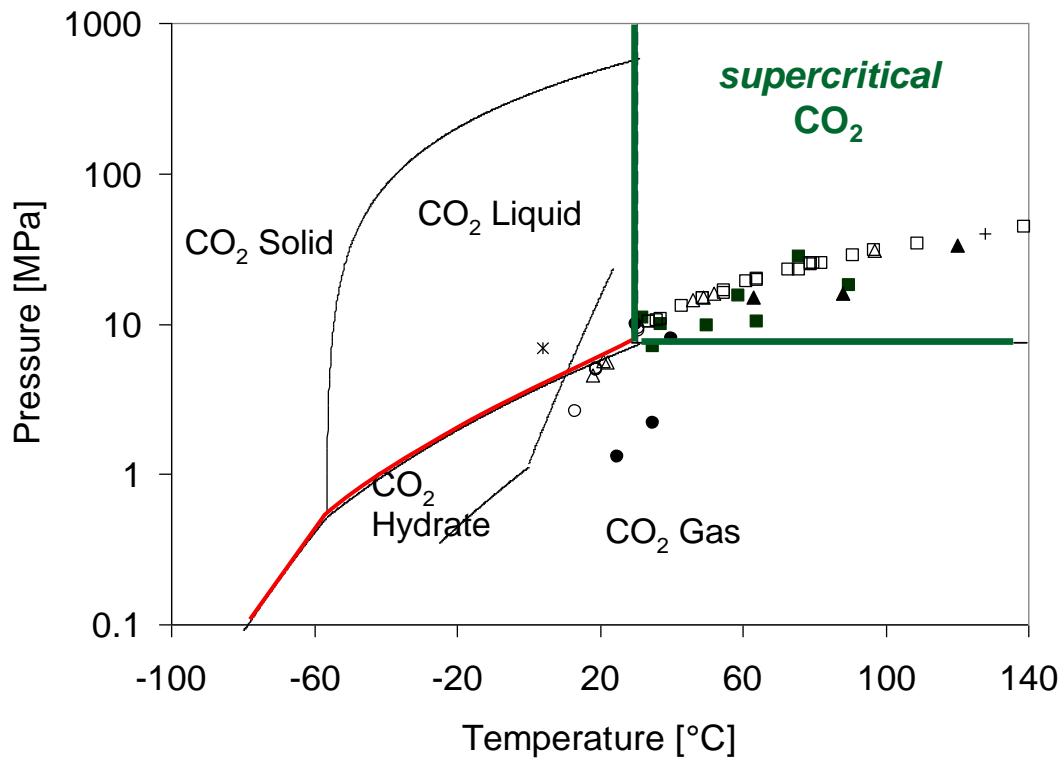
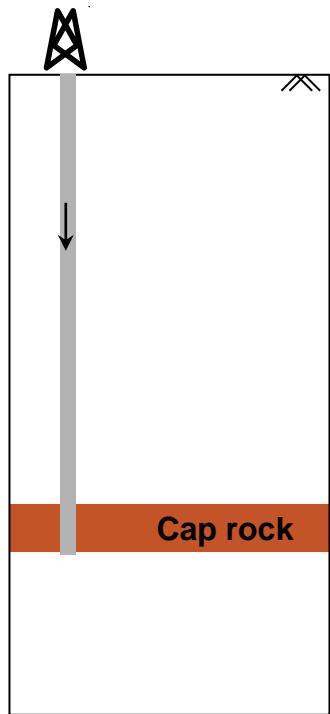
CO₂ geo-storage



Geological Storage of CO₂



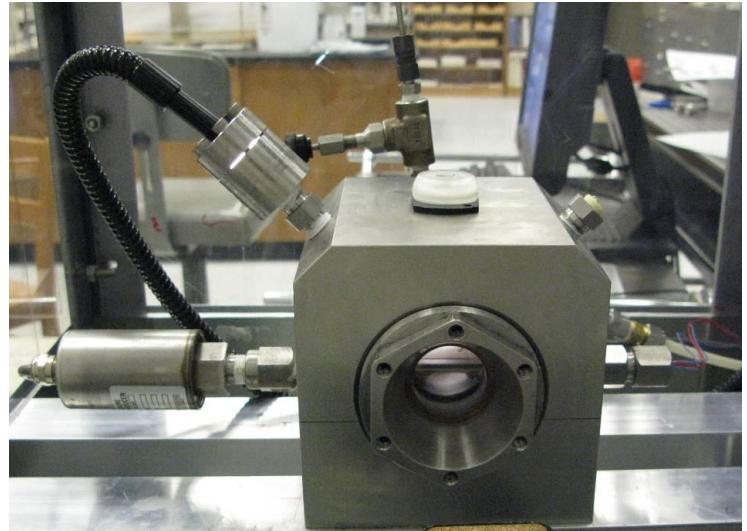
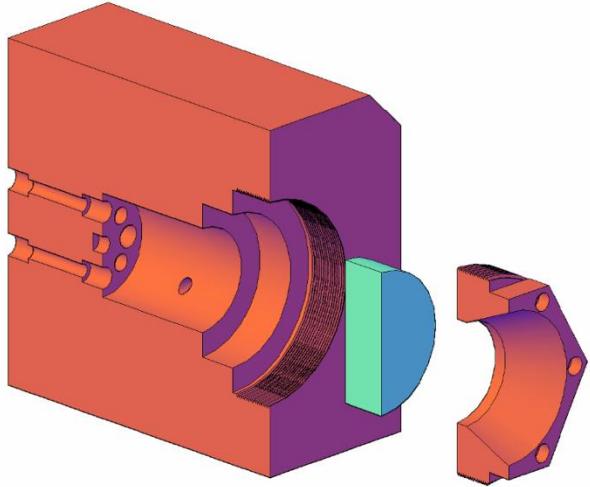
CO_2 Properties



Water and Liquid CO₂ Properties

Property	[units]	CO ₂ liquid	H ₂ O liquid
Heat capacity c_p	[kJ·kg ⁻¹ ·K ⁻¹]	2.3	4.2
Thermal cond. λ	[W·m ⁻¹ ·K ⁻¹]	~0.13	0.56
Thermal Diff. κ	[m ² s ⁻¹]	6.1×10 ⁻⁸	1.3×10 ⁻⁷
Viscosity μ	[Pa·s]	(2-to-8)×10 ⁻⁵	~1.5×10 ⁻³
Density ρ	[kg·m ⁻³]	~938-to-800	1003
Bulk Modulus	[GPa]	0.1-to-0.3	2.1-to-2.3
V_P	[m/s]	~400-to-600	1450-to-1520
Electrical cond.	[S/m]	< 0.01	f(c) - seawater: ~5
Dielectric permit.	[]	~ 1.5	~79

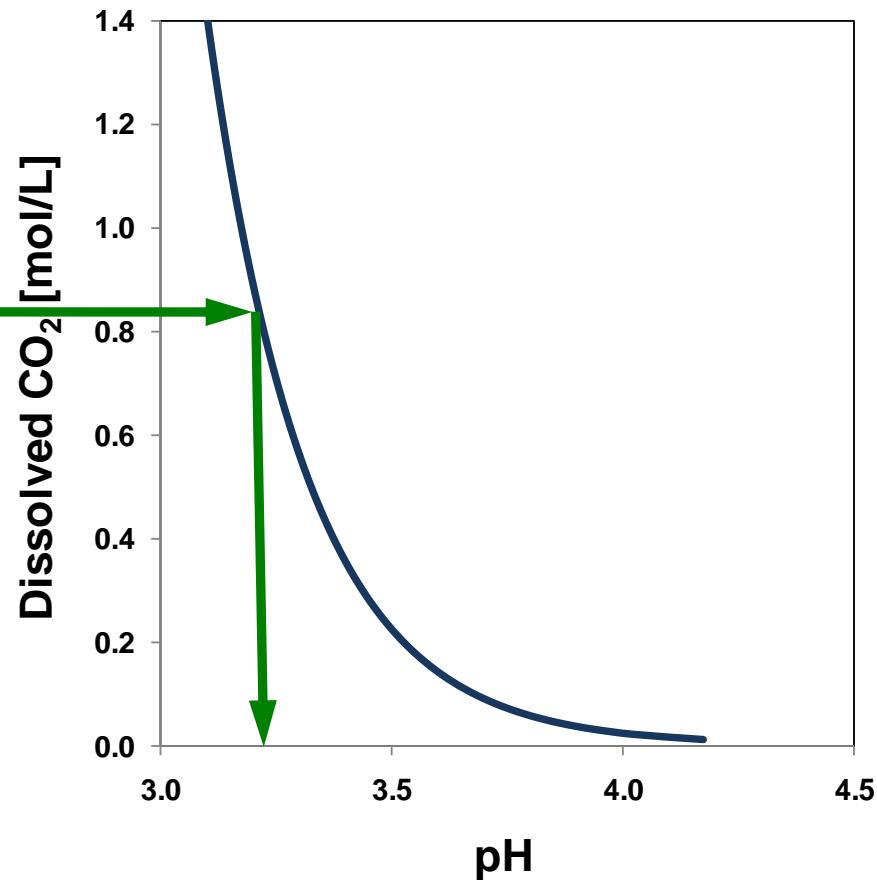
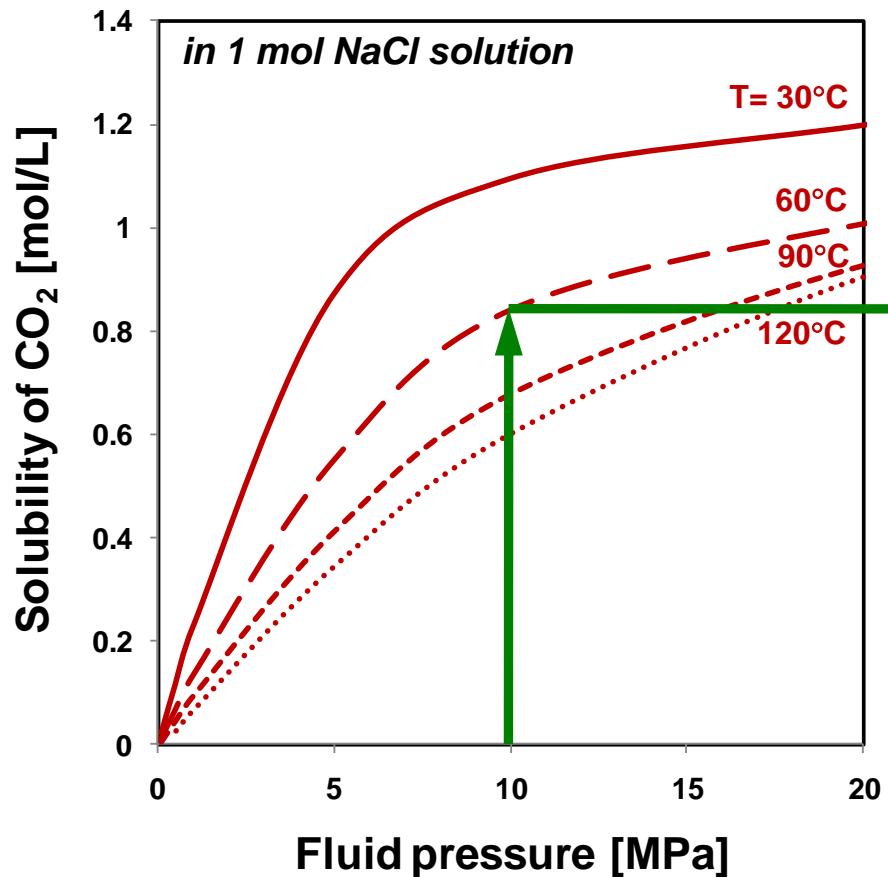
Diffusion of CO₂ in H₂O



Water diffusion into liquid CO₂: D~10⁻⁷ m²/s



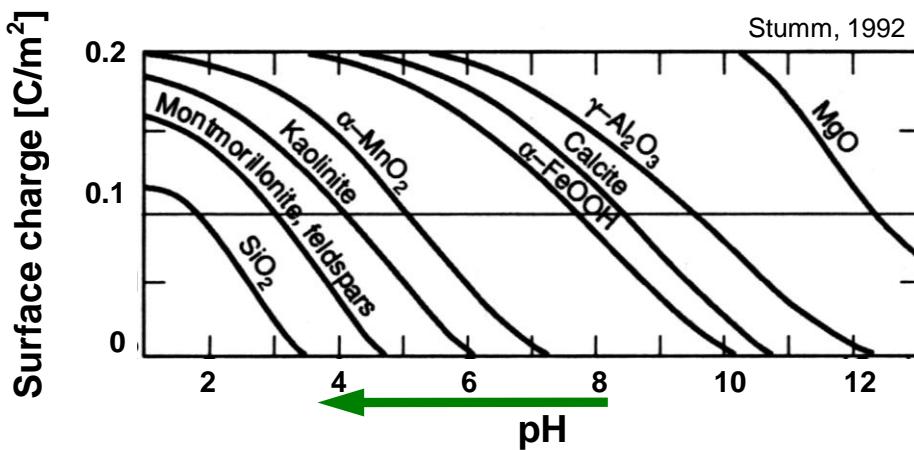
Solubility of CO₂ in Water - pH



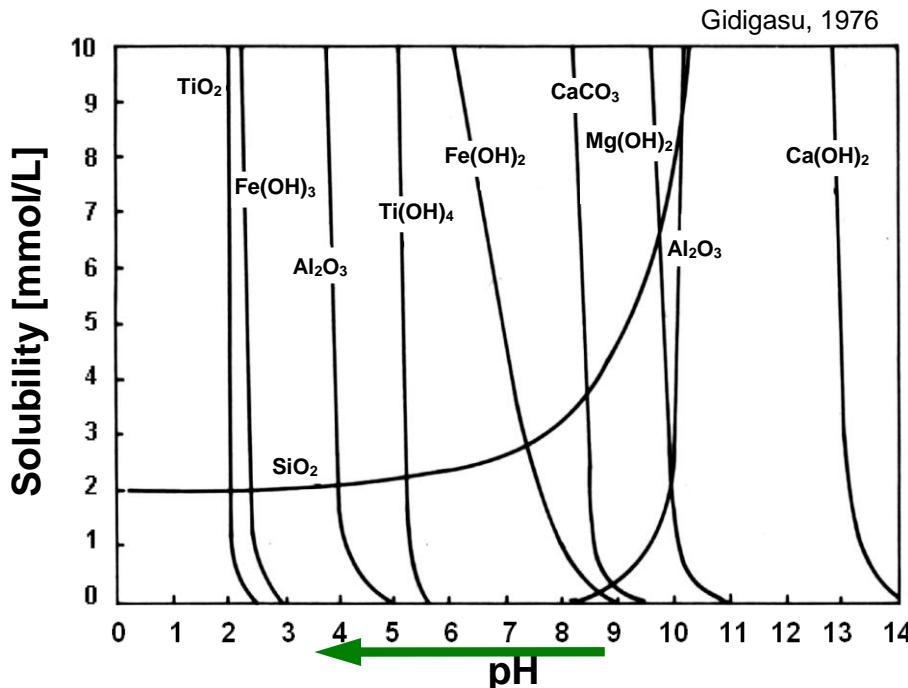
(1) change in surface charge → change in fabric

(2) mineral dissolution

pH and Minerals



→ change in fabric



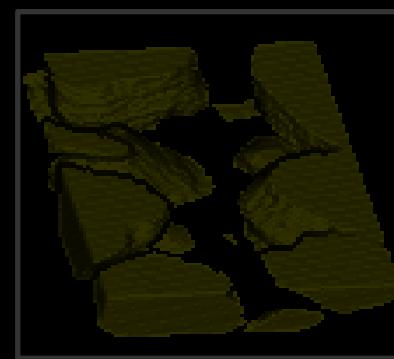
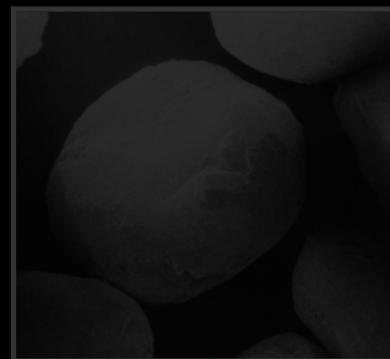
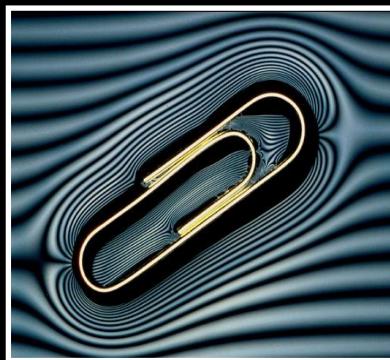
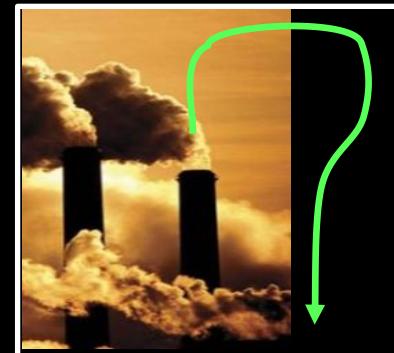
→ mineral dissolution

Summary: CO₂ Geological Storage

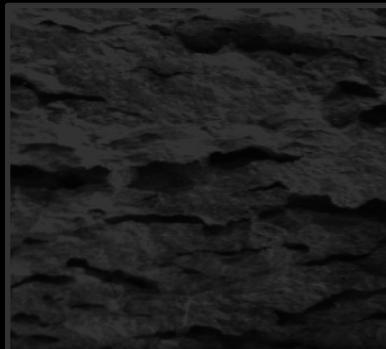
More sustainable use of fossil fuels

PT: typically in supercritical regime

- | | | |
|------------------------|---|---|
| Liquid CO ₂ | → low viscosity
→ low B, κ', σ _{el} | → invasion pattern?
→ geophysical monitoring |
| Acidifies water | → surface charge (+)
→ mineral dissolution | → clay fabric in shale cap rock?
→ stress field? permeability? |



interfaces

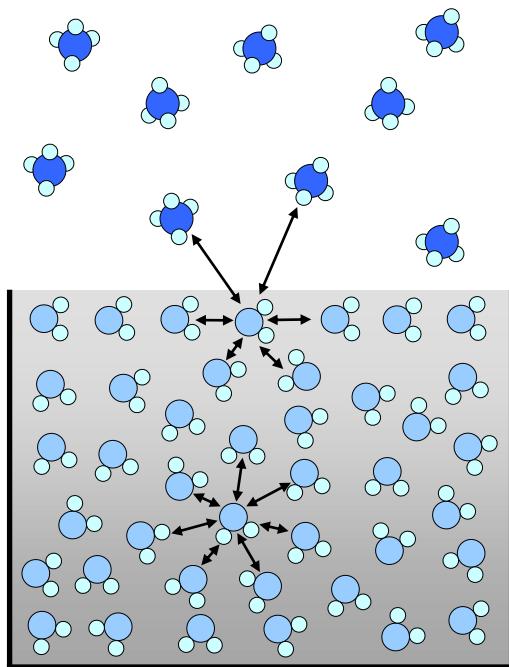


Surface Tension

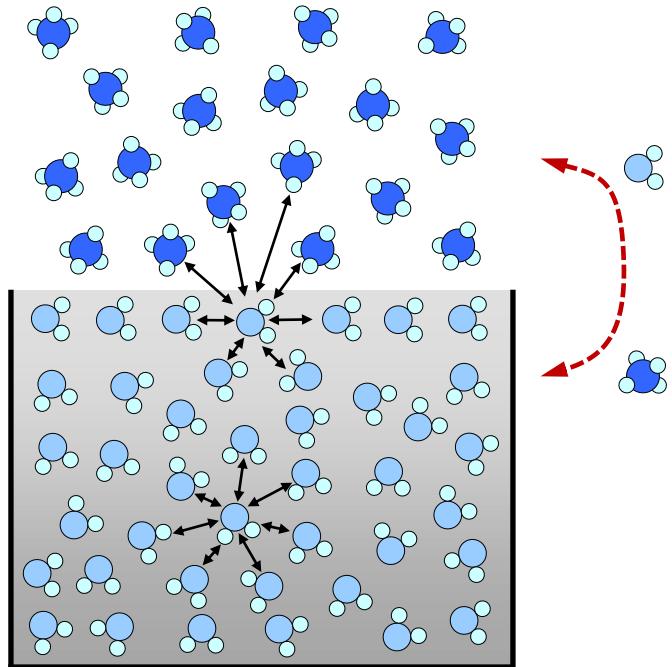


$\text{CO}_2\text{-H}_2\text{O}$: Interfacial Interaction

Low P



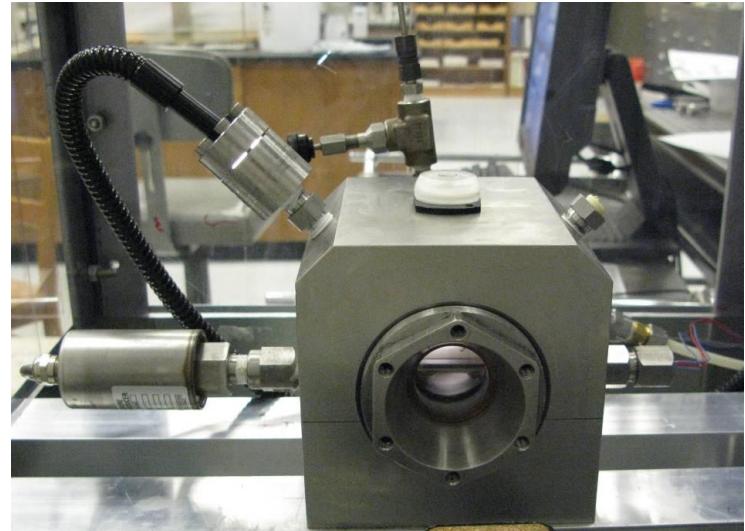
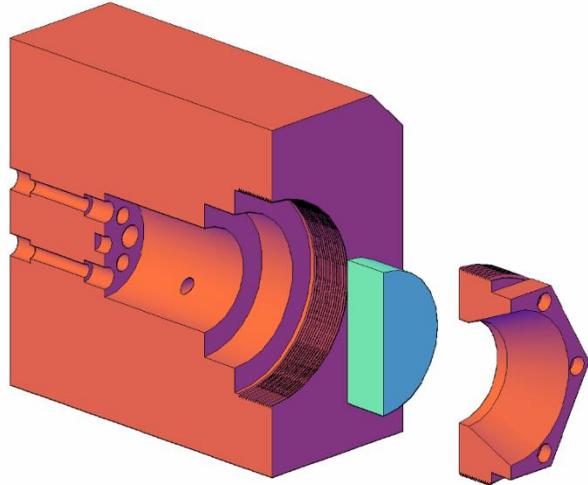
High P



(1) *mutual diffusion of $\text{CO}_2\text{-H}_2\text{O}$*

(2) *interfacial tension = $f(P)$*

Surface Tension and Contact Angle



Water droplet in

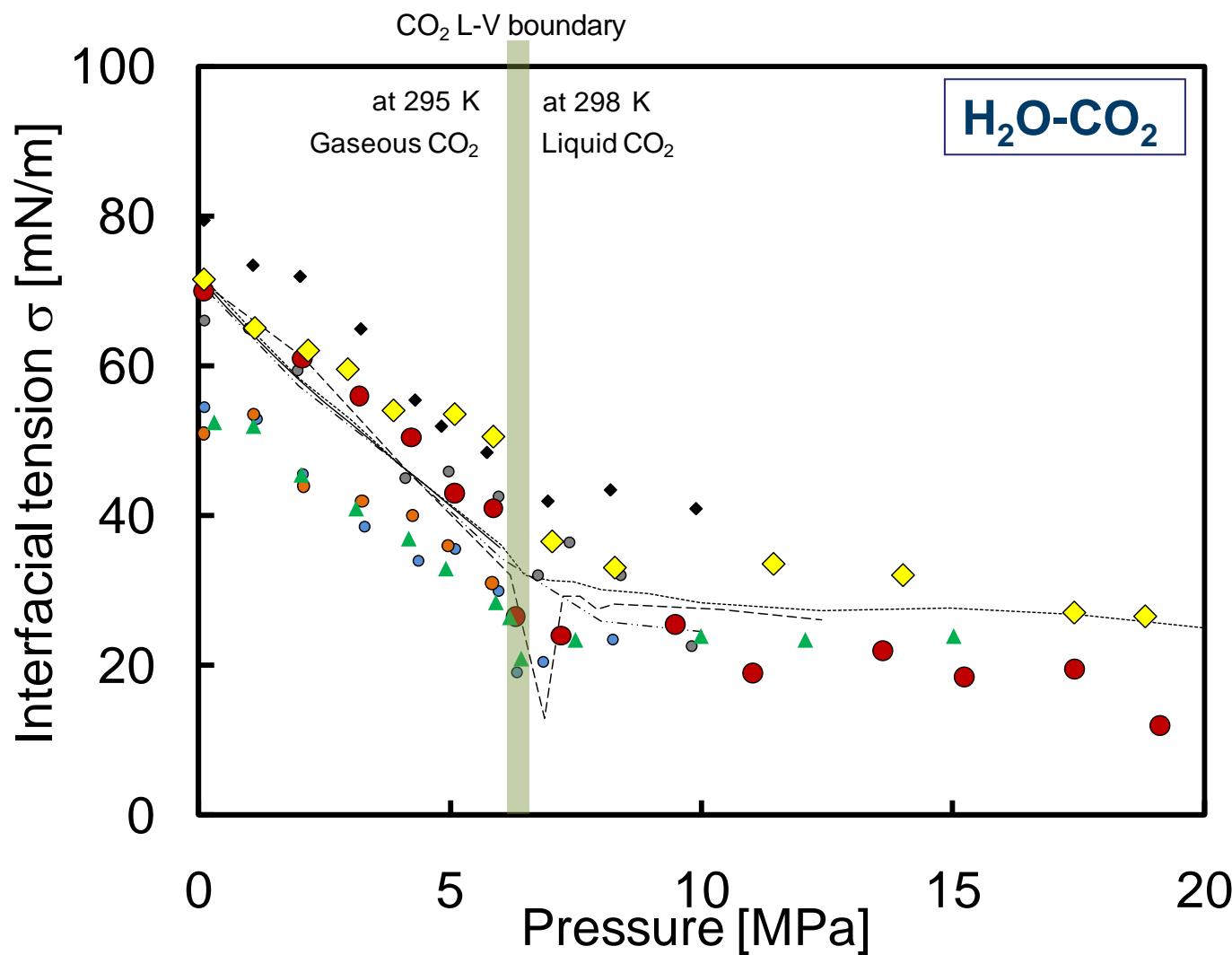
CO_2 gas



CO_2 liquid

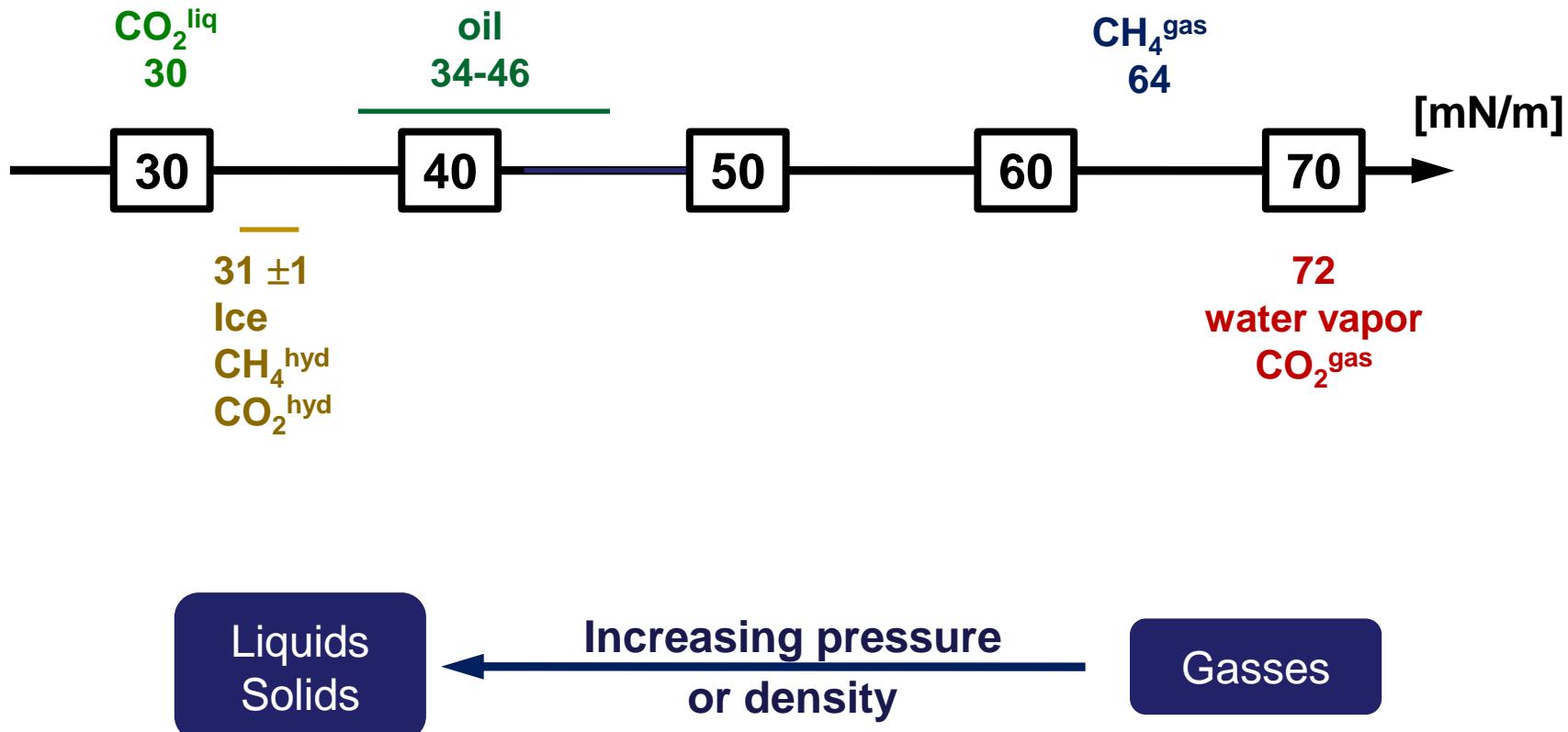


Surface Tension = f(P)



Interfacial Tension

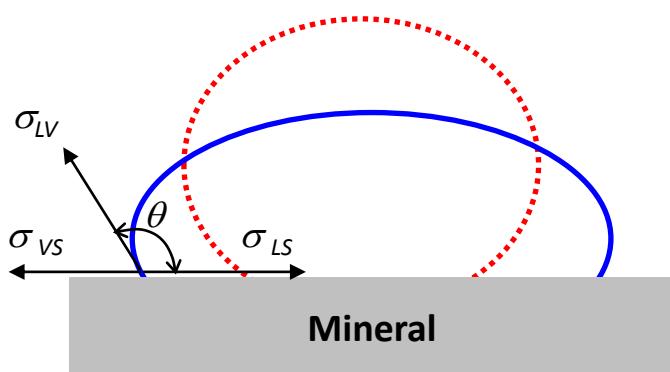
water and ...



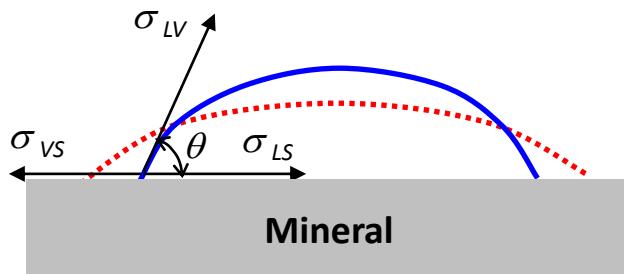
$$\cos \theta = \frac{\sigma_{VS} - \sigma_{LS}}{\sigma_{LV}}$$

Contact Angle

Non-wetting droplet



Wetting droplet

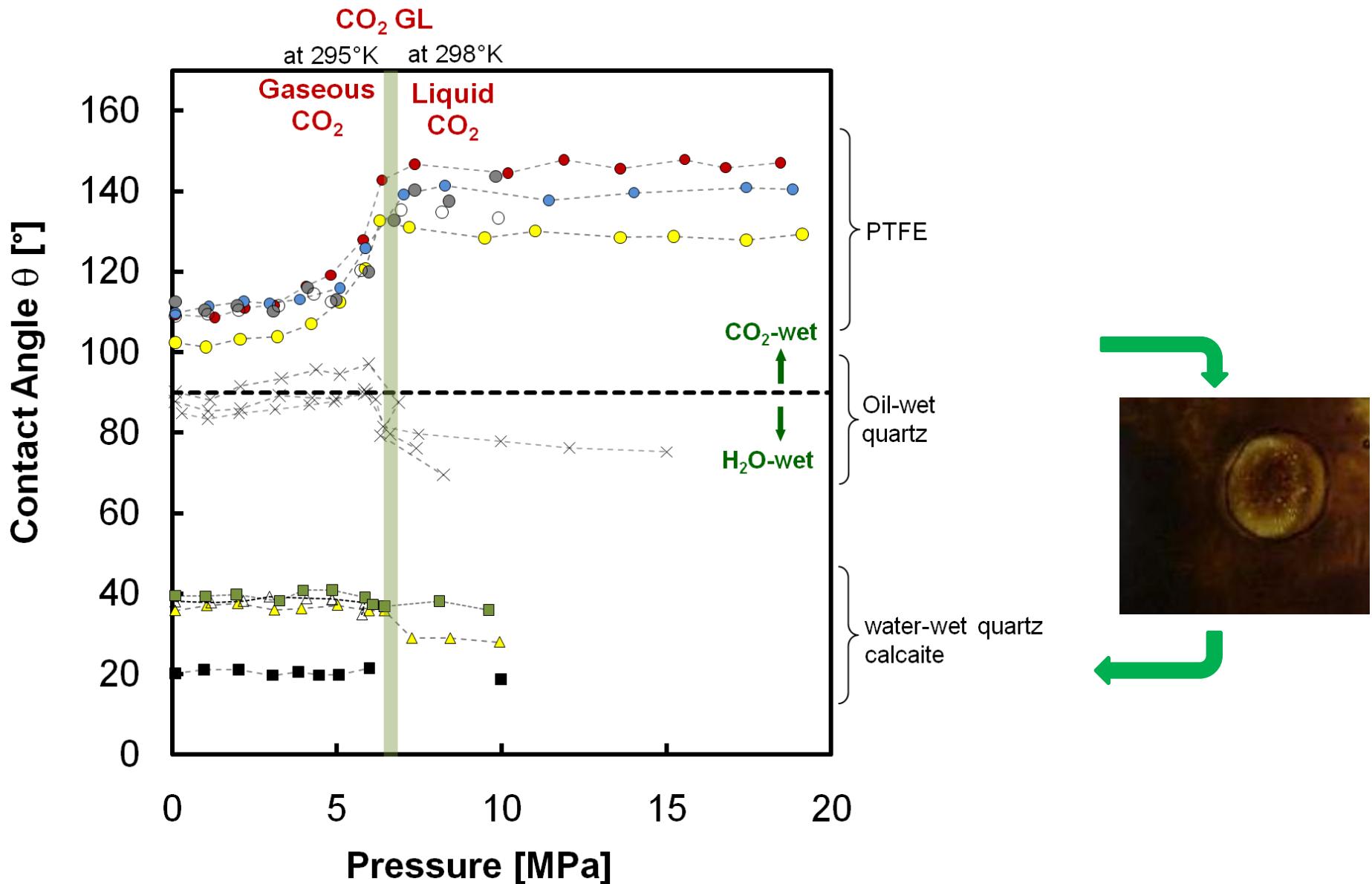


$$\sigma_{LV} \downarrow \rightarrow \theta \uparrow$$

$$\sigma_{LV} \downarrow \rightarrow \theta \downarrow$$

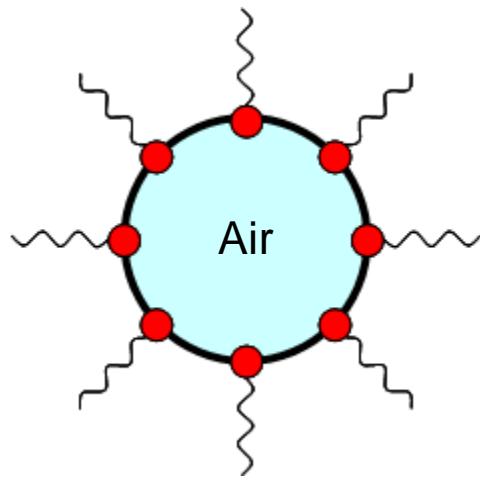
$$\cos \theta = \frac{\sigma_{VS} - \sigma_{LS}}{\sigma_{LV}}$$

Contact Angle = f(P_{gas})

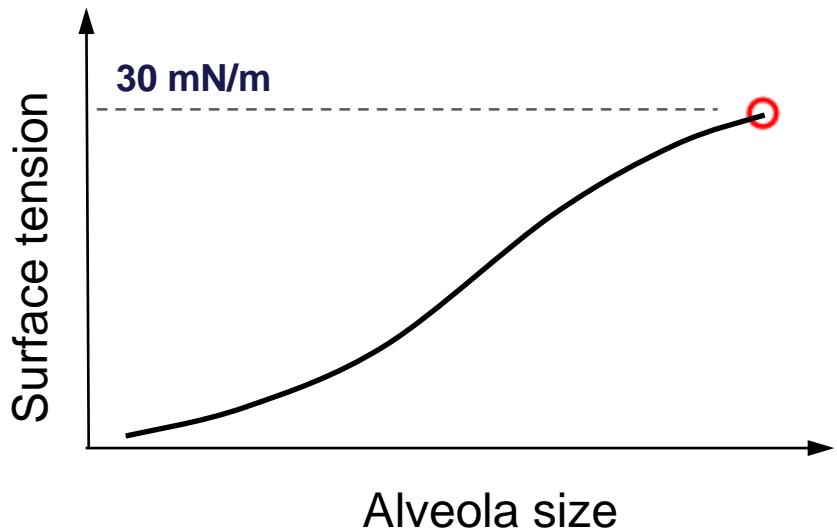


Other Effects - Surfactants

Pulmonary self-regulation:



hydrophobic ● hydrophilic



Surfactant → Surface tension = $f(\text{pore size})$ → S-u data interpretation

Capillary Pressure - Laplace



$$(P_{nw} - P_w) \pi R^2 = 2\pi R \sigma \cos \theta$$

$$\Delta P = \frac{2\sigma}{R} \cos \theta$$

$$\Delta P = \frac{2\sigma}{R} \cos \theta \quad \xrightarrow{\Delta P = \frac{\rho}{M} RT \ln\left(\frac{1}{h_r}\right)} R = 2 \sigma_{LV} \cos \theta \frac{M}{\rho RT \ln\left(\frac{1}{h_r}\right)}$$

$$\Delta P = \sum_m (T_m - T) \quad \xrightarrow{\Delta P = 2\sigma_{wi} \cos \theta \frac{1}{\sum_m (T_m - T)}} R = 2 \sigma_{wi} \cos \theta \frac{1}{\sum_m (T_m - T)}$$

Characteristic curves Δu -S for:

water-gas
water-ice

water-oil
water-hydrate

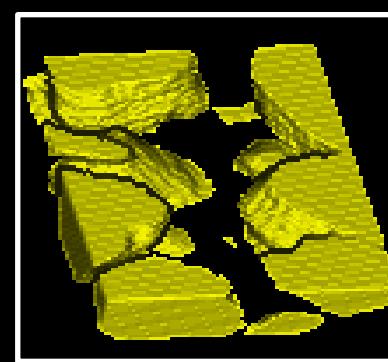
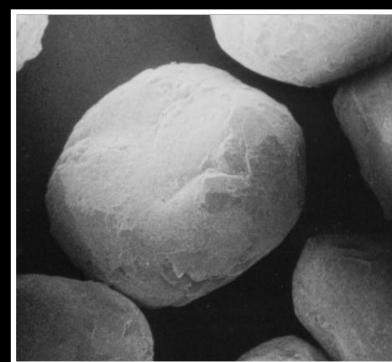
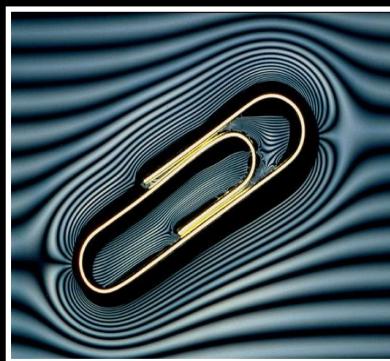
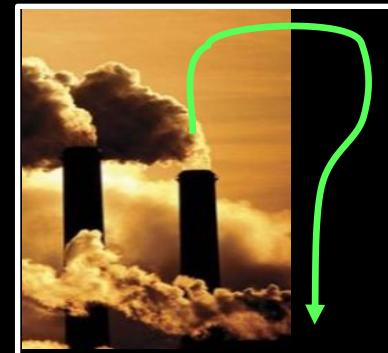
gas-oil

Summary: Interfacial Phenomena

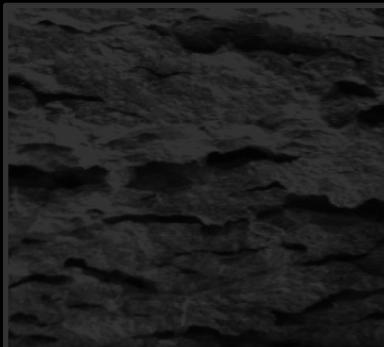
Interfacial tension: participating molecules and PT (or density)

Contact angle: varies with interfacial tensions

Implications: capillarity
characteristic curves
resource recovery

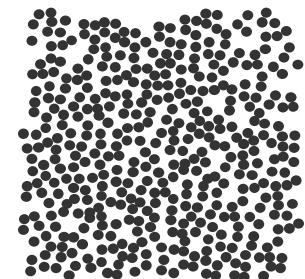
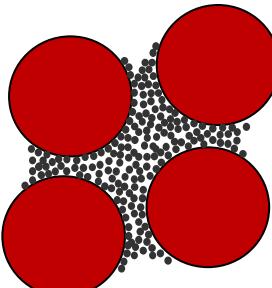
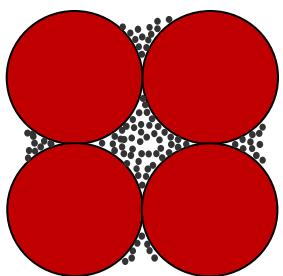
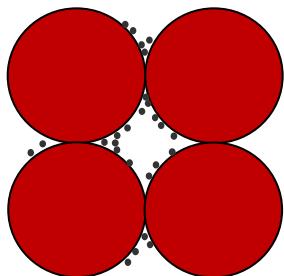
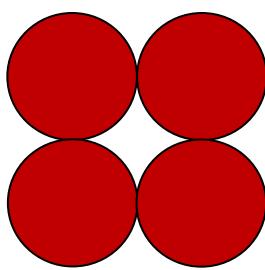


grains and pores



Grain Size Distribution: The Role of Fines

Critical fines
Content FC*
(for mechanical properties ...)

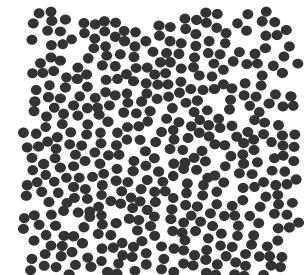
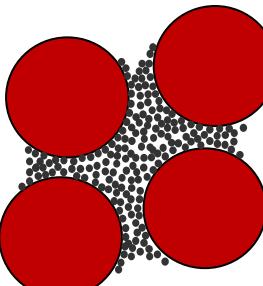
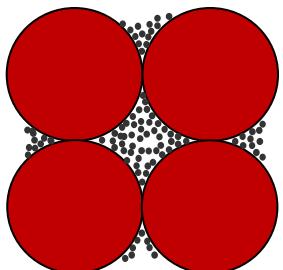
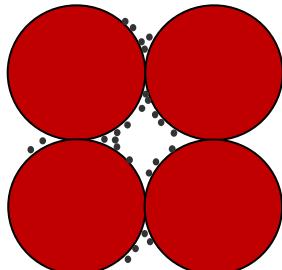
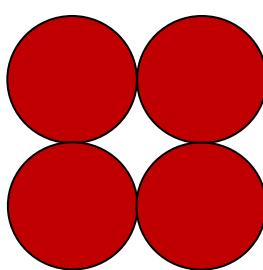


$$FC^* = \frac{M_{\text{fine}}}{M_{\text{total}}} = \frac{e_{\text{coarse}}}{1 + e_{\text{coarse}} + e_{\text{fine}}}$$

Sediment	$e_{1\text{kPa}}$	FC*
Silt	~0.7	~ 25 %
Kaolinite	~1.5	~ 20 %
Illite	~3.7	~ 11 %
Montmorillonite	~5.4	~ 8 %

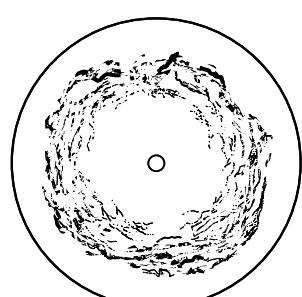
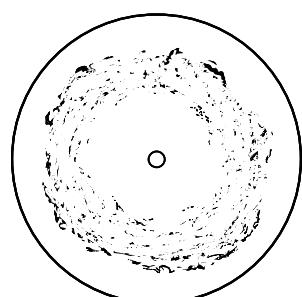
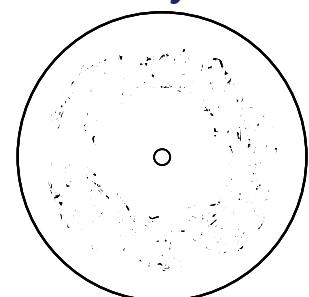
Fines Migration and Clogging

Critical fines
Content FC*

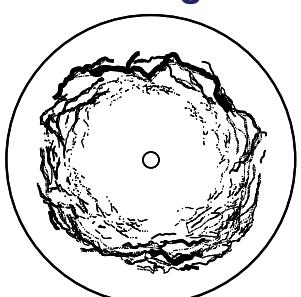


*fines migration
& clogging*

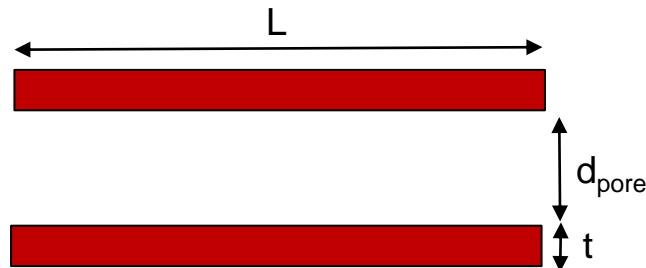
early Q



after large Q



Grains and Pores: Clays



$$d_{\text{pore}} = \frac{2 e}{S_s \rho}$$

MEAN PORE SIZE

Sediment compaction

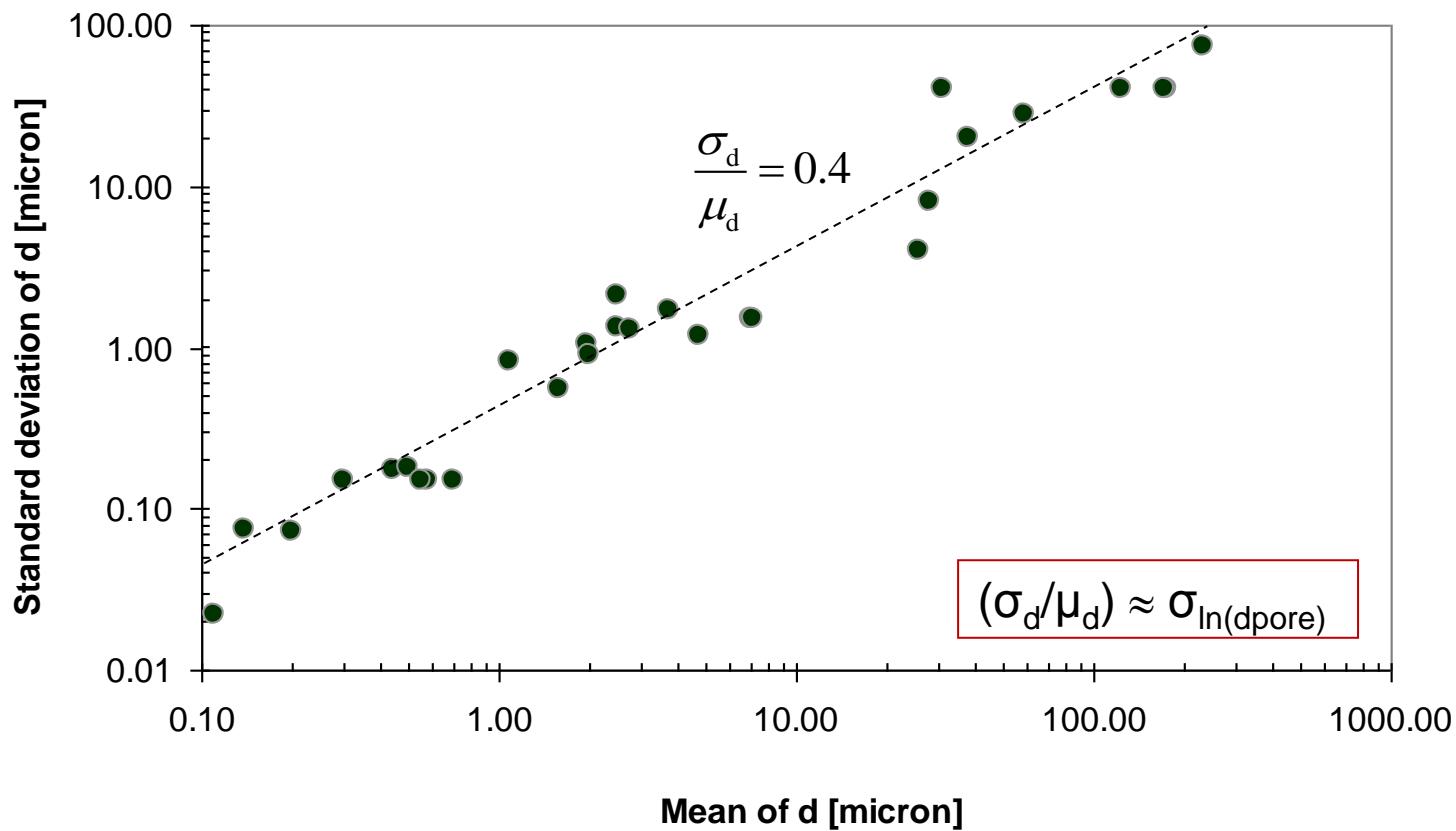
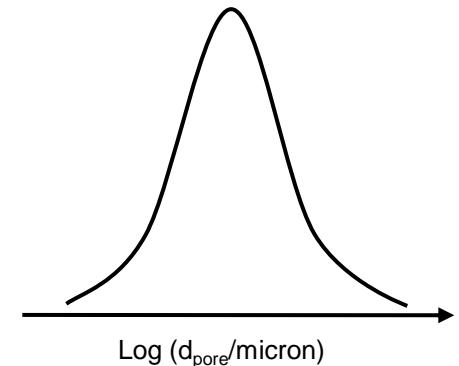
$$e = e_{1\text{kPa}} - C_c \log\left(\frac{\sigma'}{1\text{kPa}}\right)$$

Sediment	$e_{1\text{kPa}}$	C_c	$S [\text{m}^2/\text{g}]$	mean d_{pore}	$\Delta P [\text{Mpa}]$
Silt	~0.7	0.02-0.09	0.045-1	5 μm	0.05
Kaolinite	~1.5	0.19-0.3	10-20	0.5 μm	0.5
Illite	~3.7	0.5-1.1	65-100	0.05 μm	5
Montmorillonite	~5.4	1-2.6	300-780	0.005 μm	50

@ $\sigma'=100$ kPa

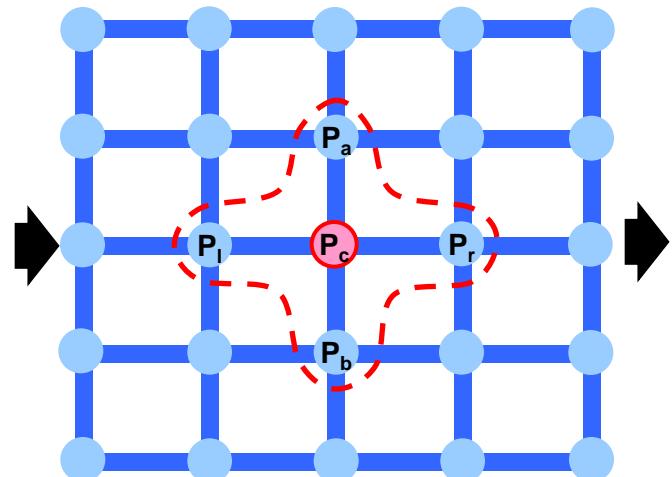
$\sigma_{LV}=70$ mN/m

Pore Size Distribution



Network Models – Upscaling

Poiseuille's Eq.
$$q = \frac{\pi R^4}{8\eta \Delta L} \Delta P \quad \left(\alpha = \frac{\pi R^4}{8\eta \Delta L} \right)$$



Mass Balance at Nodes

$$0 = \sum q_c$$

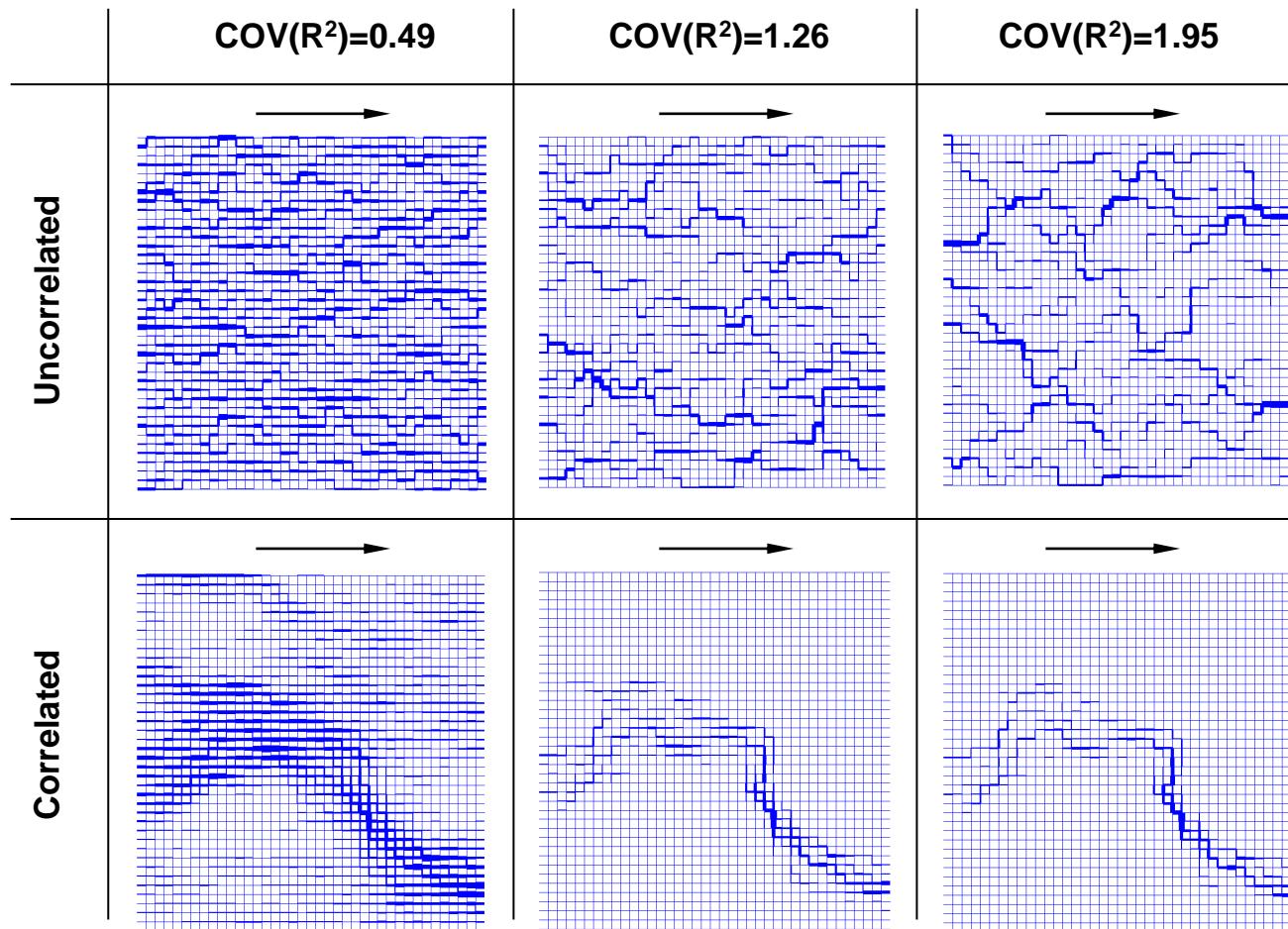
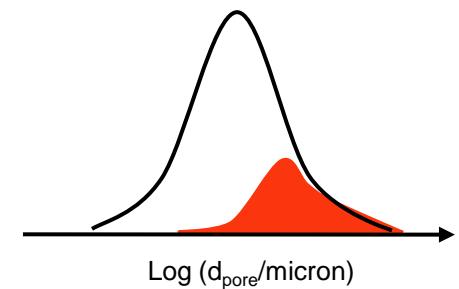
$$0 = \alpha_a (P_a - P_c) + \alpha_b (P_b - P_c) + \alpha_r (P_r - P_c) + \alpha_l (P_l - P_c)$$

$$P_c = \frac{\alpha_a P_a + \alpha_b P_b + \alpha_r P_r + \alpha_l P_l}{(\alpha_a + \alpha_b + \alpha_r + \alpha_l)}$$

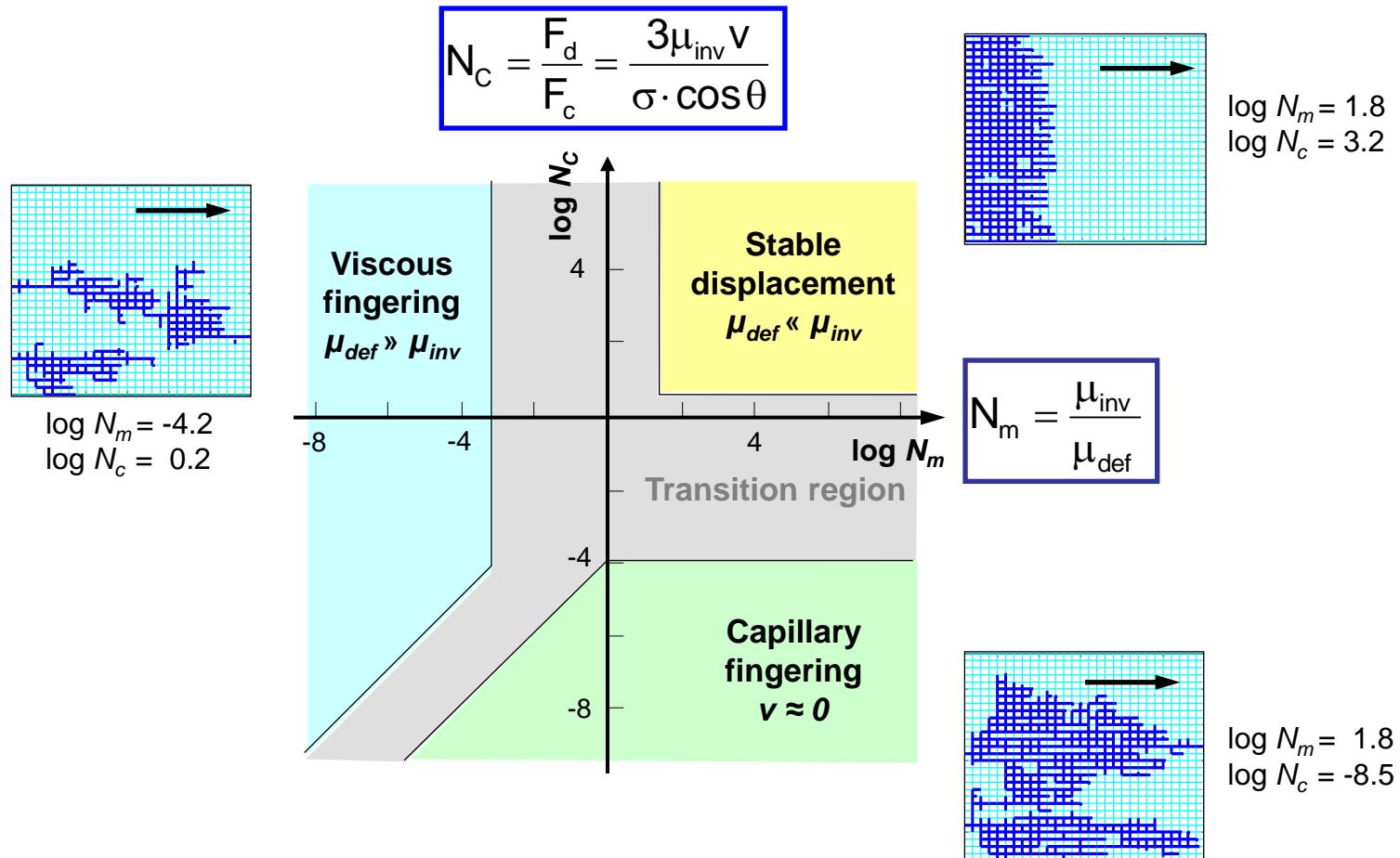
System of Equations

$$\underline{B} = \underline{\underline{A}} \underline{P} \quad \text{then} \quad \underline{P} = \underline{\underline{A}}^{-1} \underline{B}$$

Spatially Correlated Porosity

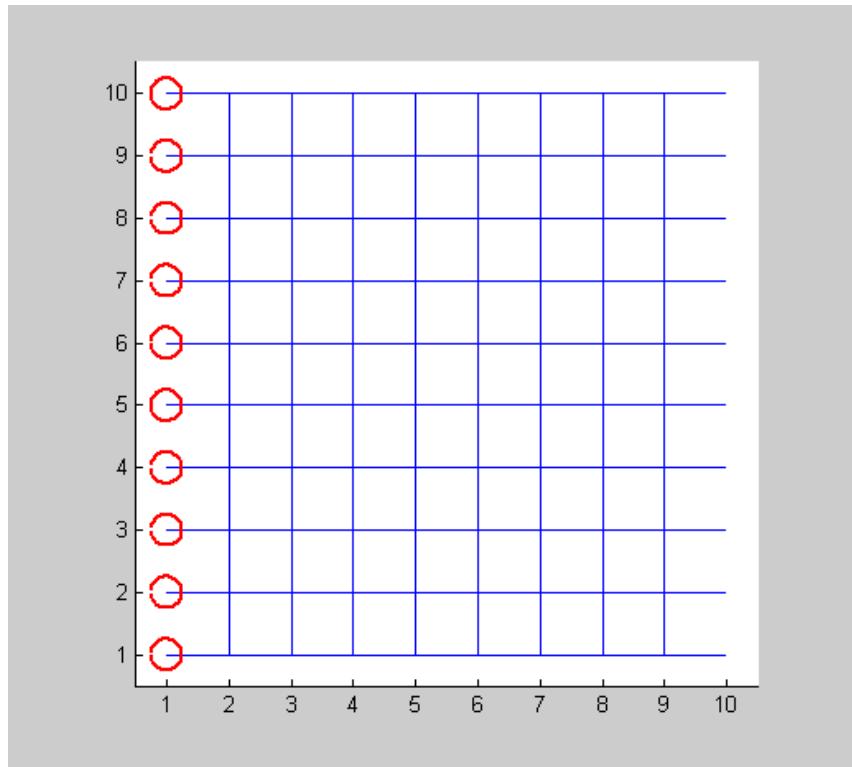


Immiscible Fluids: Invading vs. Defending

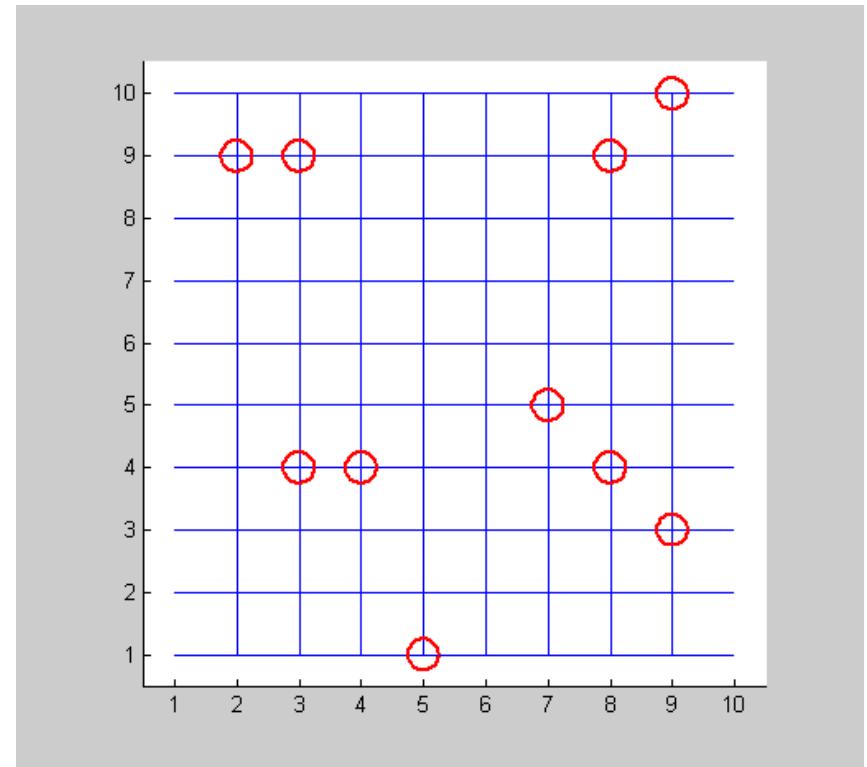


Gas Invasion vs. & Gas Nucleation

invasion

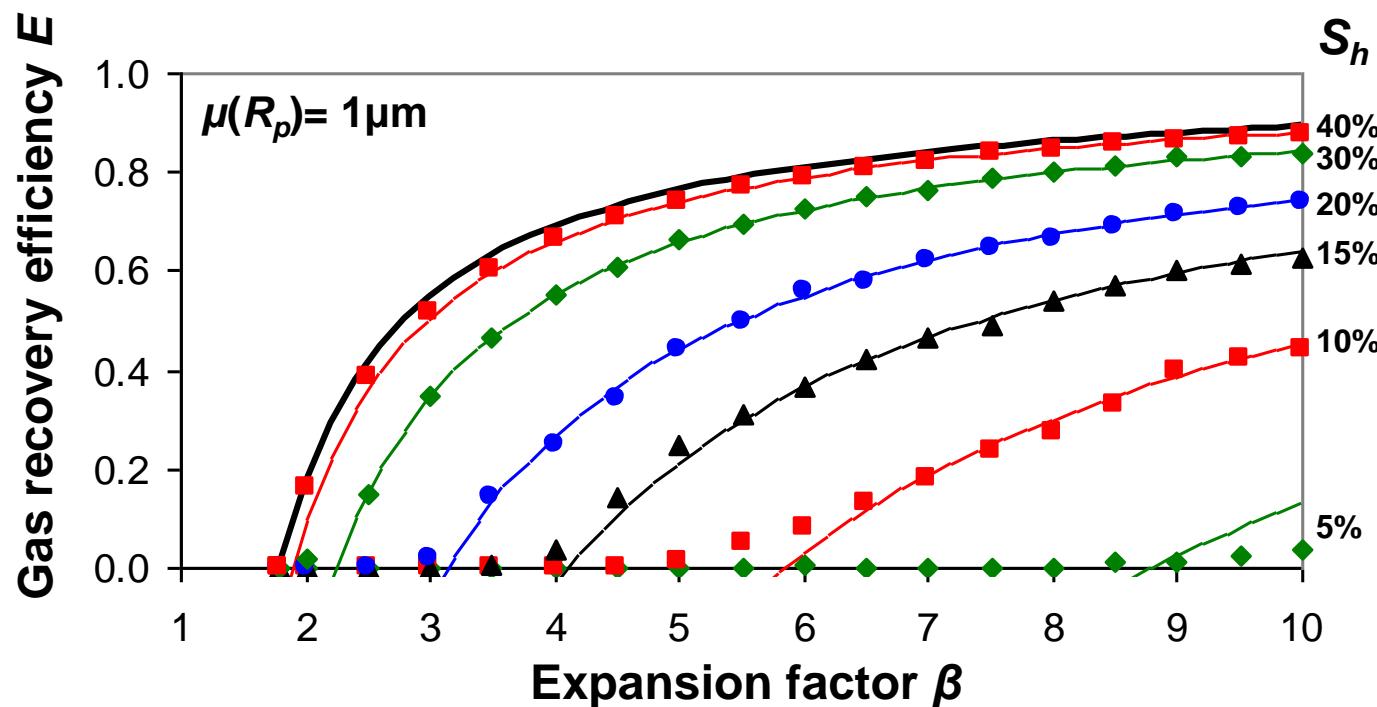
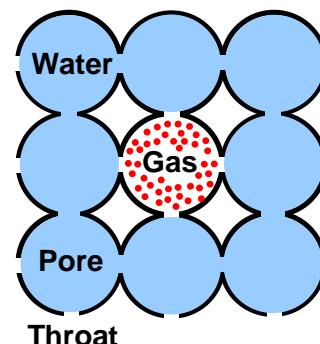


nucleation



Characteristic Curve → Recovery Efficiency

Methane Hydrates



Summary: Grains and Pores

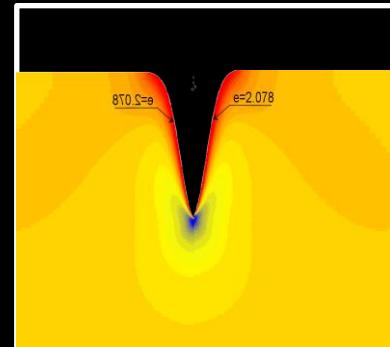
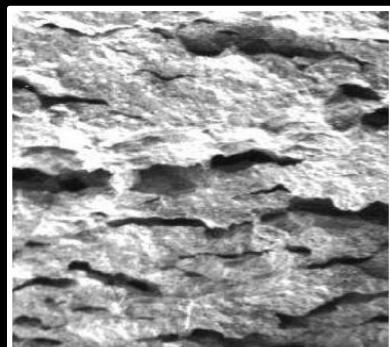
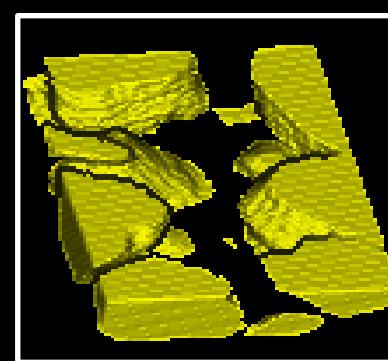
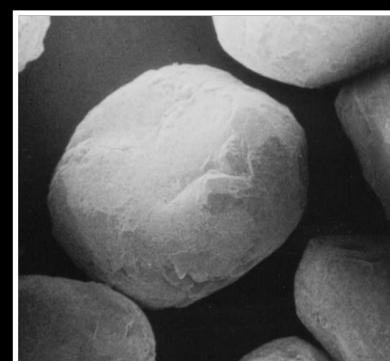
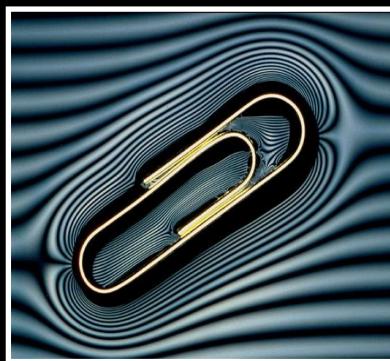
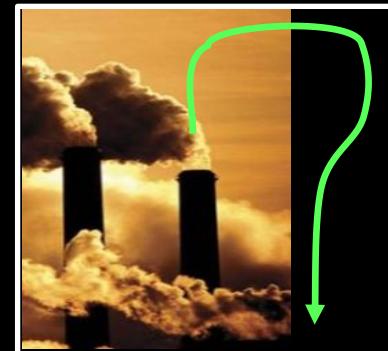
CAUTION **fines content !**

Invasion **pore size distribution and connectivity**
patterns: flow velocity and rel. viscosities

Gas invasion ≠ Gas nucleation

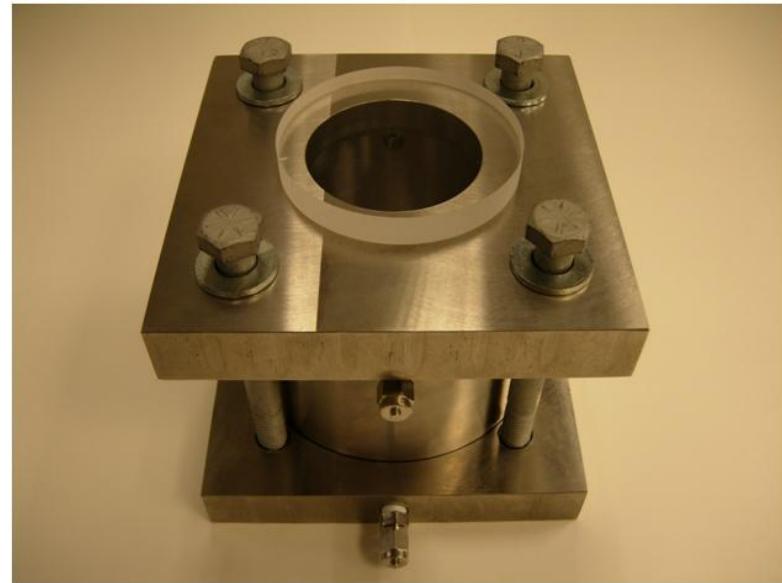
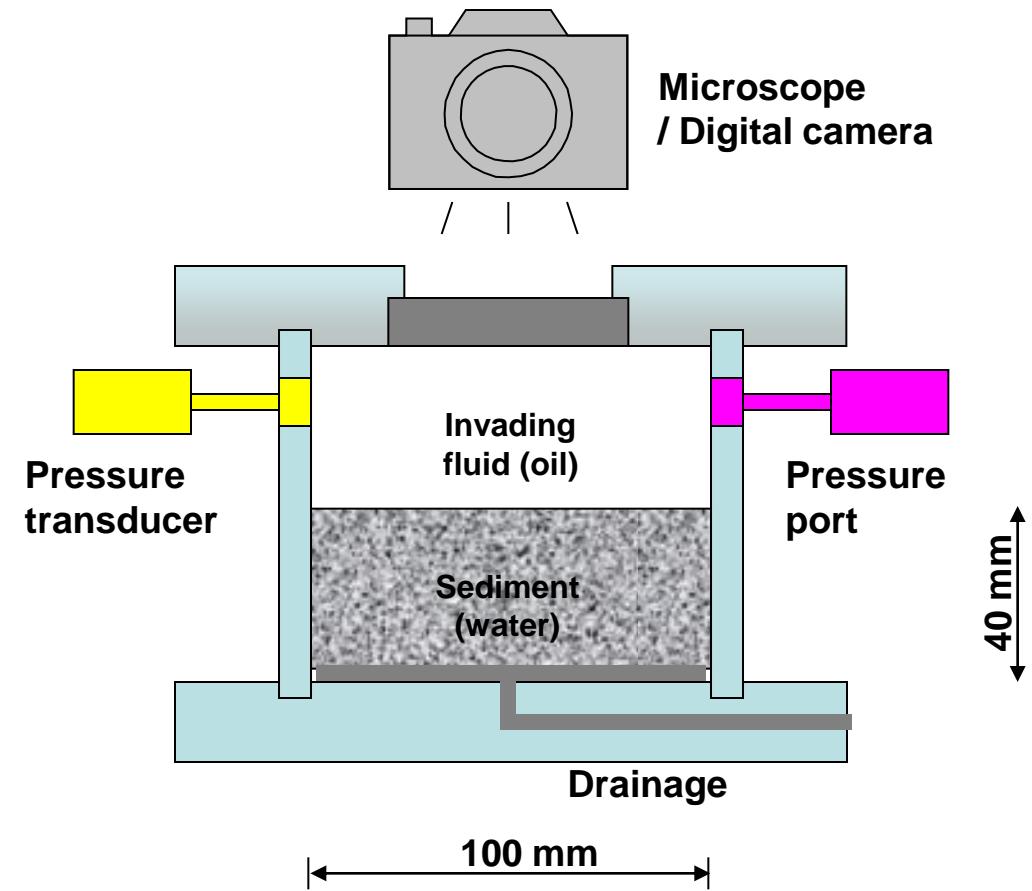
similar characteristic curve
different relative permeabilities

Characteristic curve → Recovery efficiency

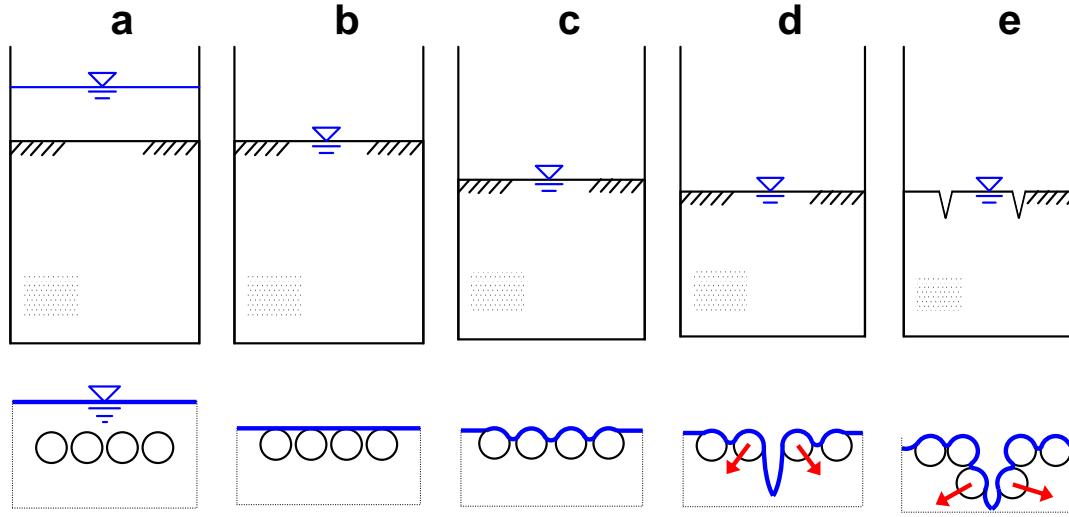
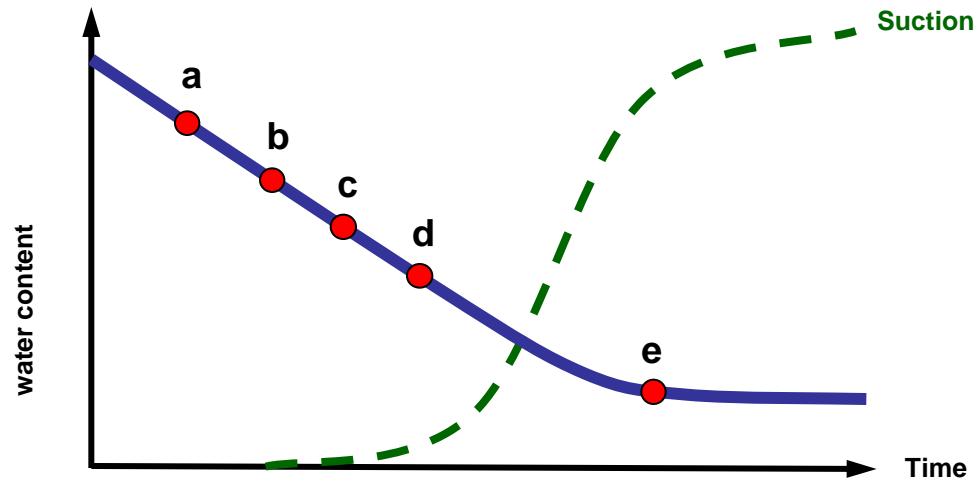


positive feedback in coupled THCM processes → localizations

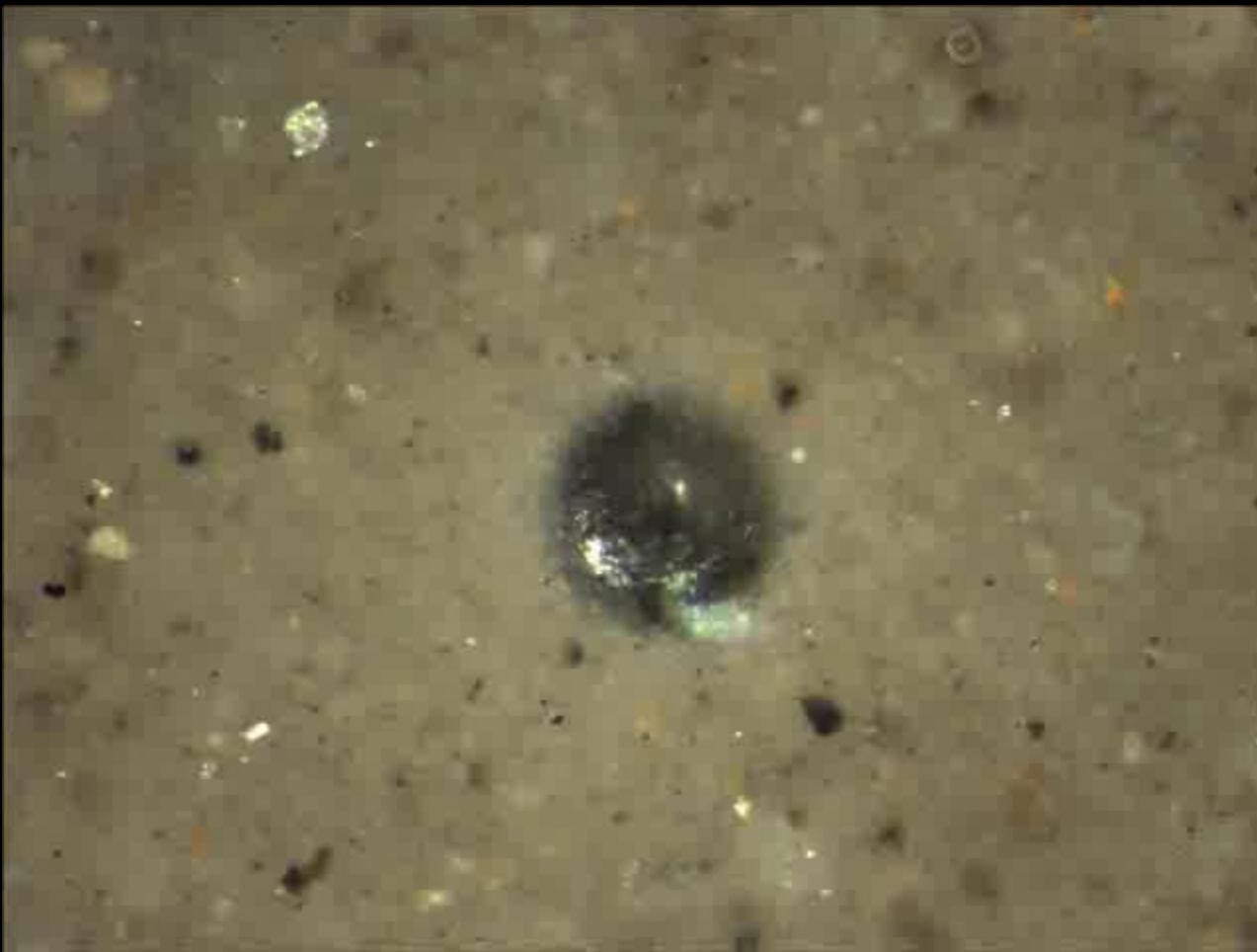
Forced Immiscible Fluid Invasion - Device



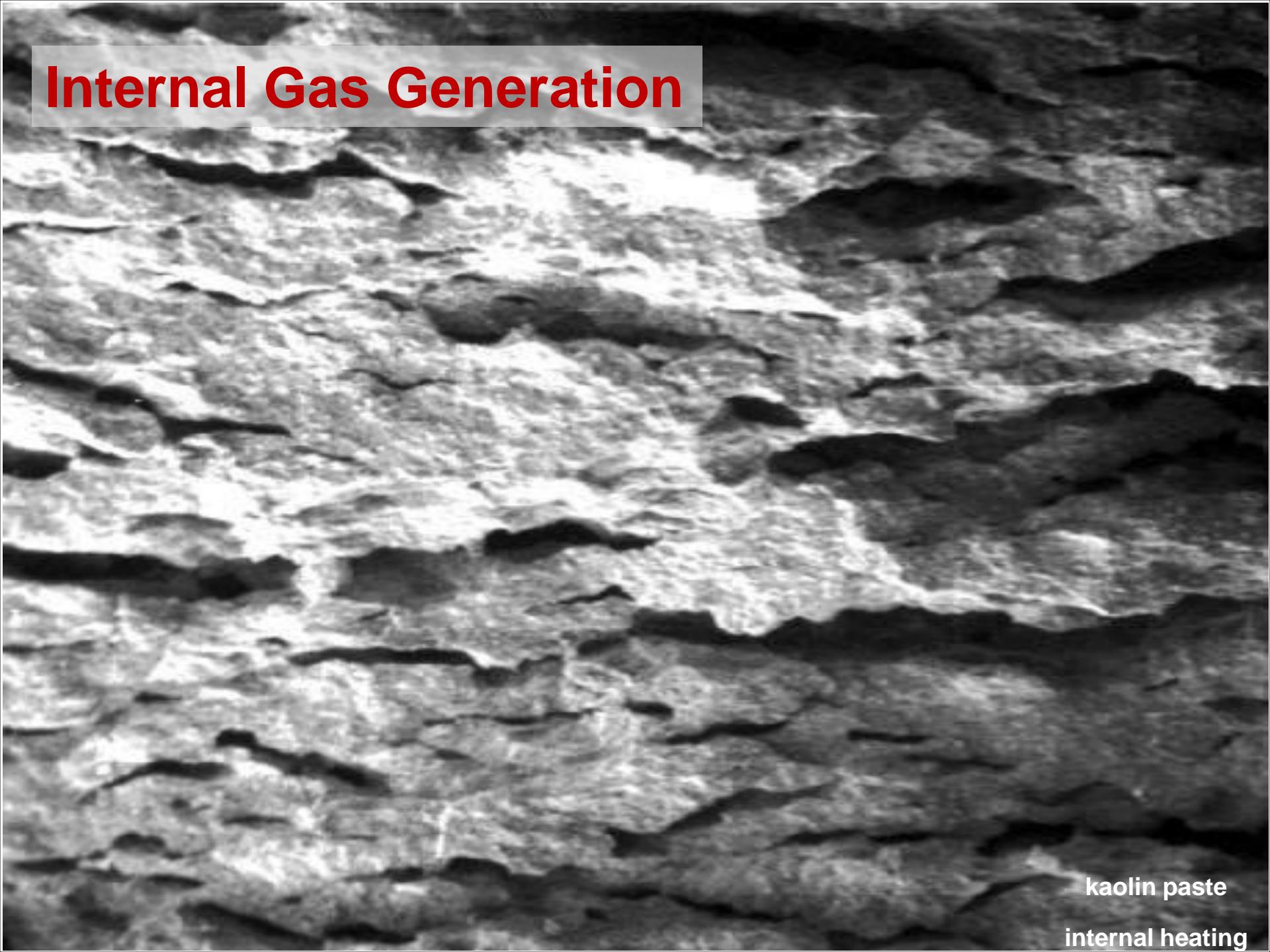
Evolution



Forcing Gas Into Sediment



Internal Gas Generation

A scanning electron micrograph (SEM) showing a porous, granular structure. The image is in grayscale, with bright, irregularly shaped regions representing pores or gas-filled spaces within a darker, more uniform matrix. The overall texture is somewhat wavy and undulating.

kaolin paste

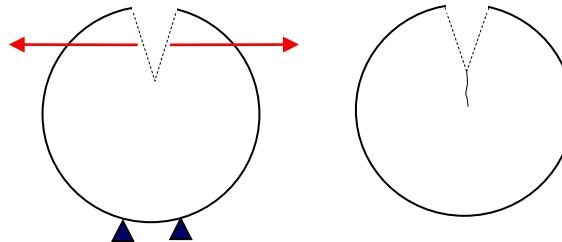
internal heating

Ice Lenses

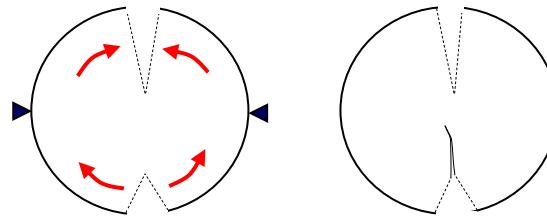


Ice Lens Formation Under Stress Boundary

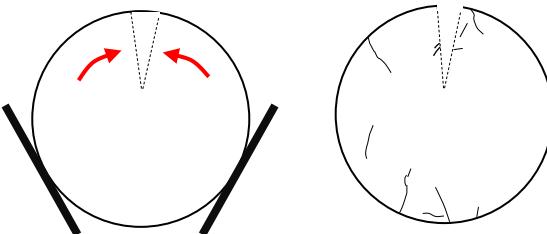
Tension



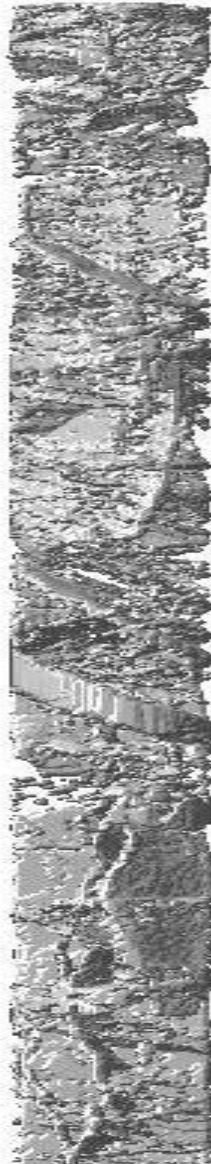
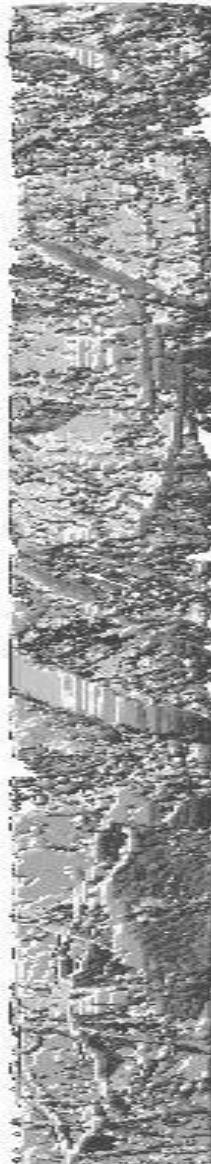
Tension / compression



Compression



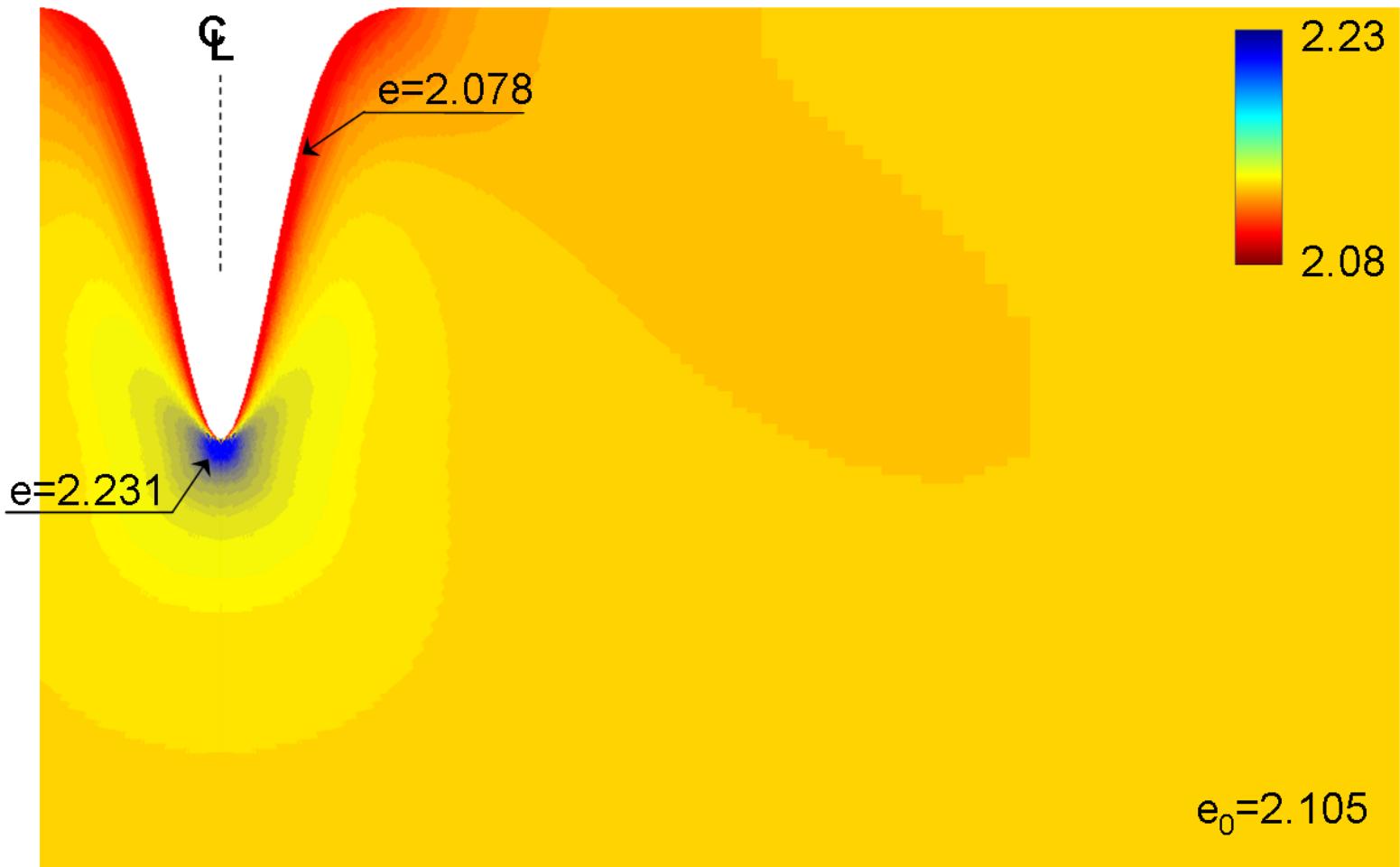
Hydrate lenses – Pressure cores



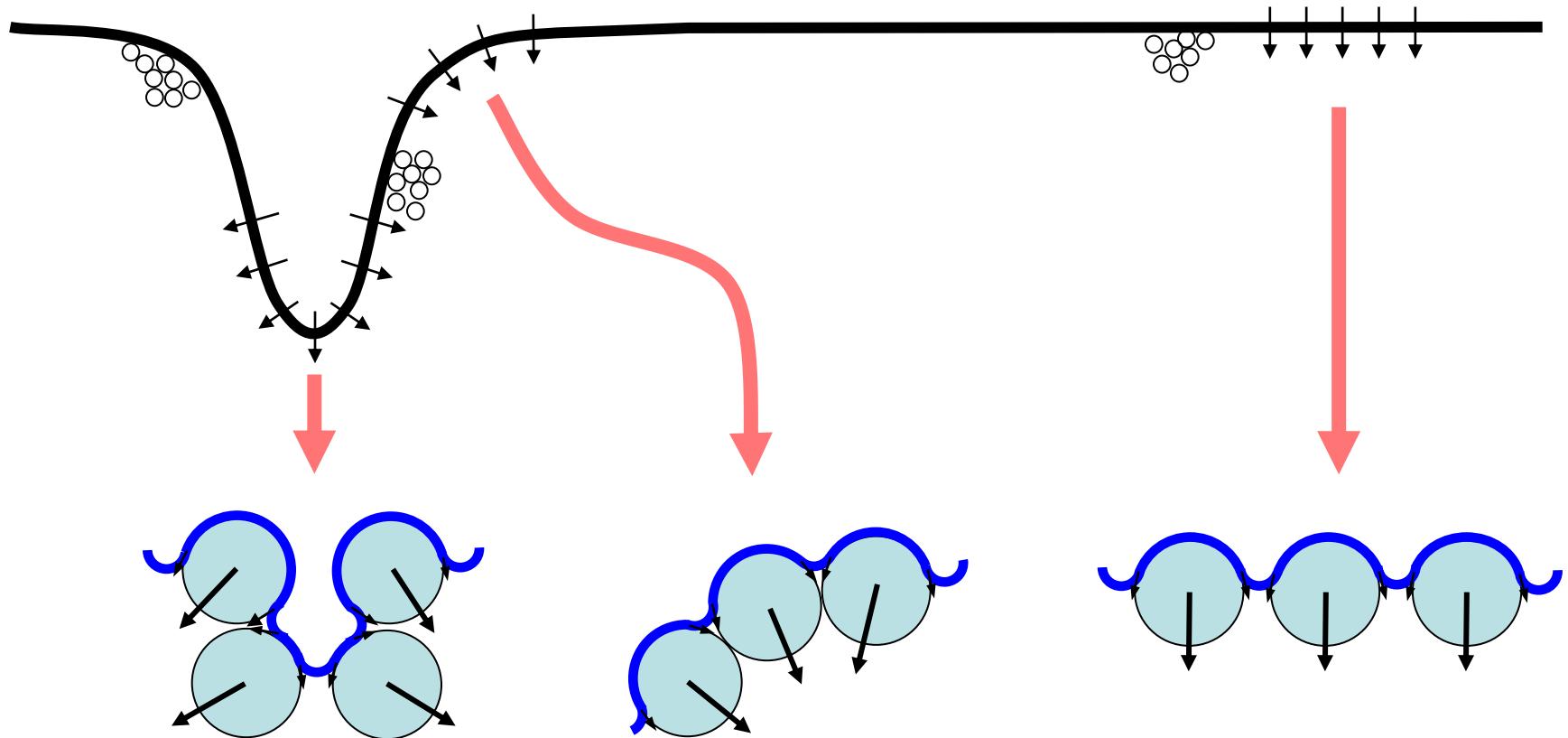
Percussion Core - KUB

Tip Conditions

$$\Delta P = \frac{\rho \sigma_s S_s}{e}$$



Gas-Driven Fracture



Invasion vs. Localization

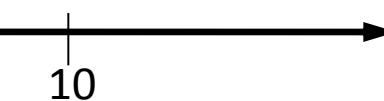
INVASION
Fluid invasion
Crystal growth in pores
Hyd.: *patchy saturation*



coarse grained soils
high effective stress

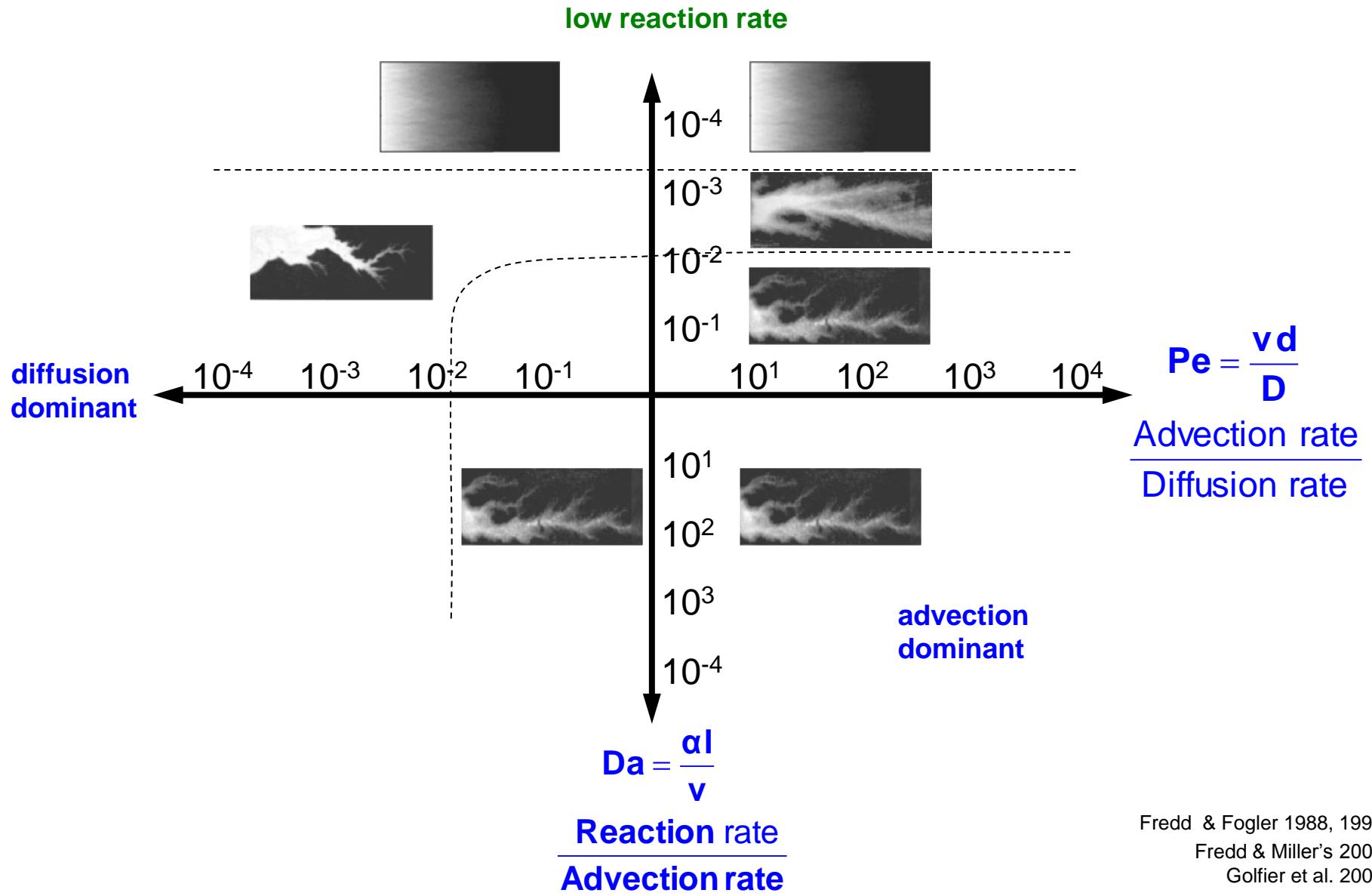
$$\frac{F_c}{N} = \frac{2\pi\sigma_{LV}}{\sigma'd}$$

LOCALIZATION
Lenses
Fractures
Hyd.: *lenses*



fine grained soils
low effective stress

Acidification: Reactive Transport

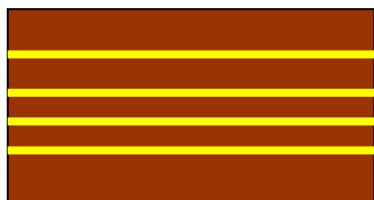


Fredd & Fogler 1988, 1998

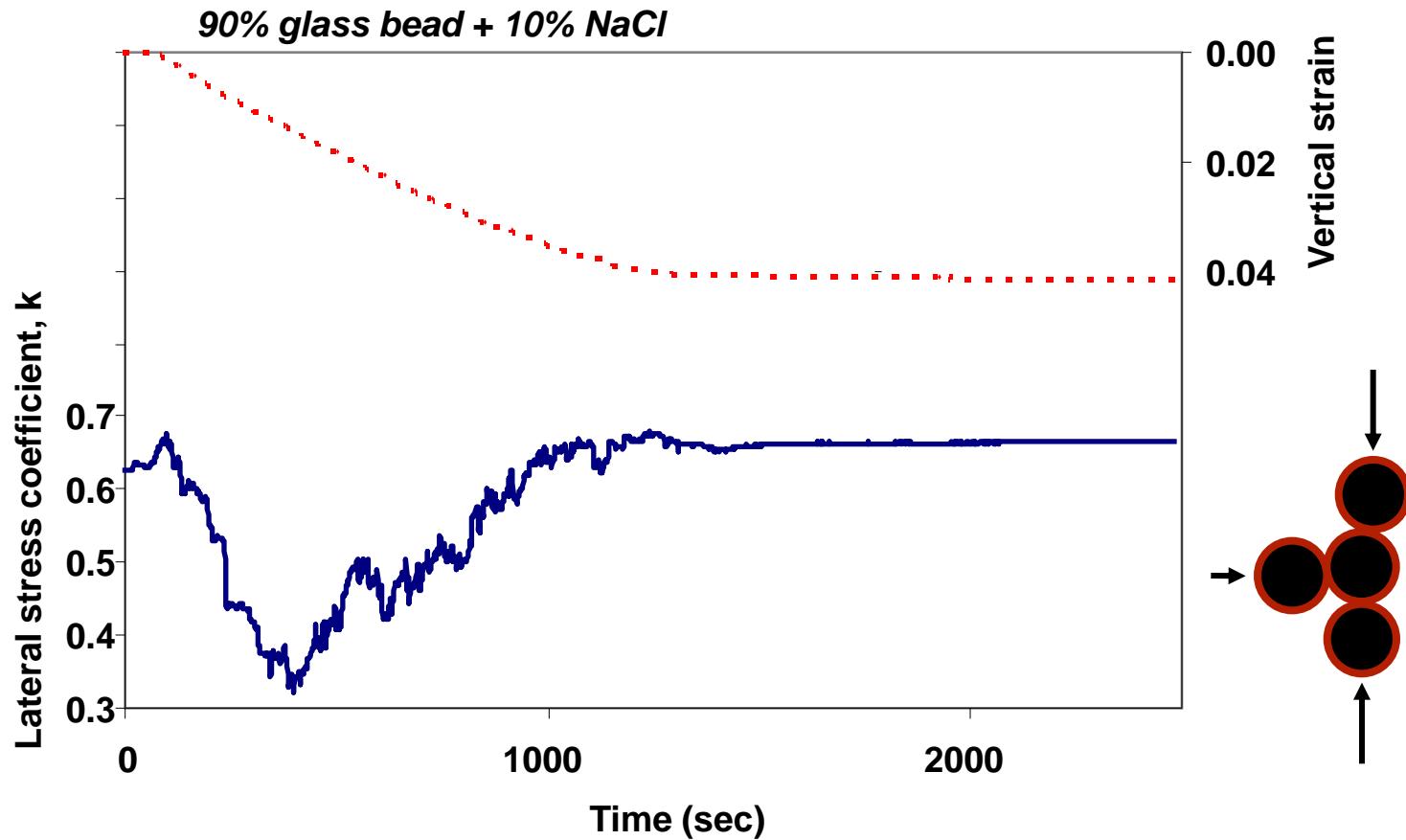
Fredd & Miller's 2000

Golfier et al. 2002

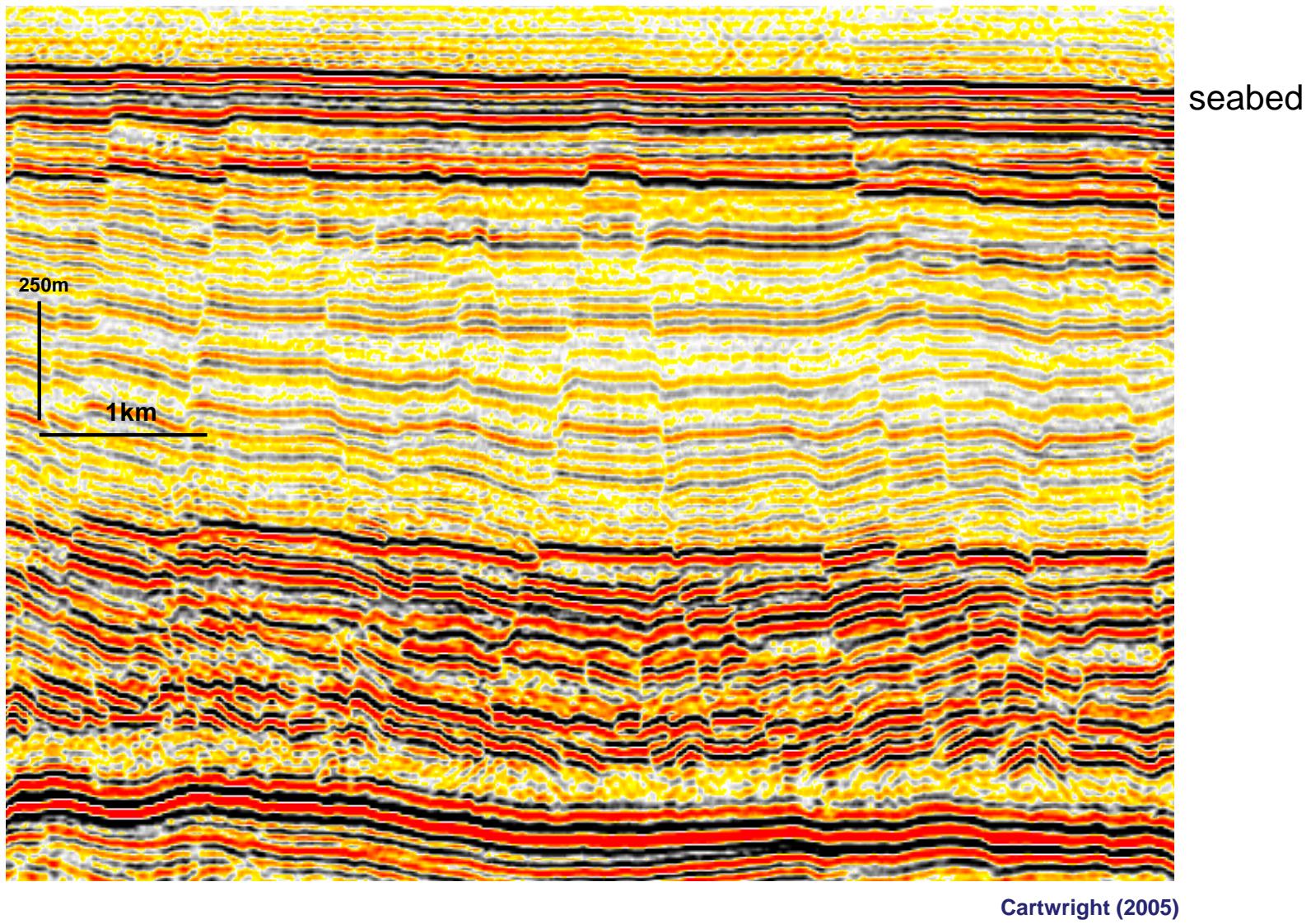
Dissolution + $k \uparrow$ + $q \uparrow$: Localized Flow



Dissolution $\rightarrow k_o \downarrow$



$k_o \rightarrow k_a$: Possible Shear Localization



Summary: Localizations

F_c/N explains hydrate formation habit

Various possible positive-feedbacks in coupled THCM

Fluid-driven fractures: desiccation cracks, gas or oil-driven

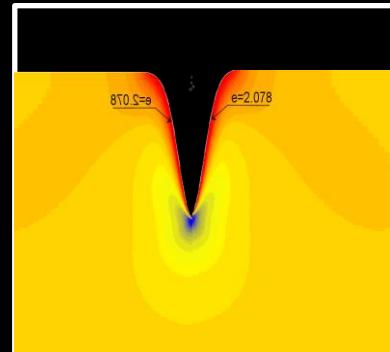
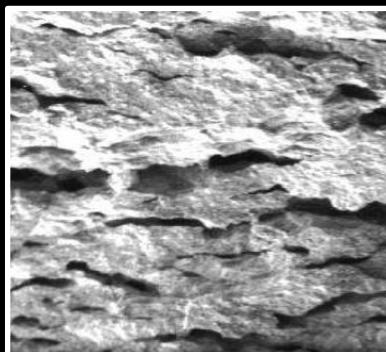
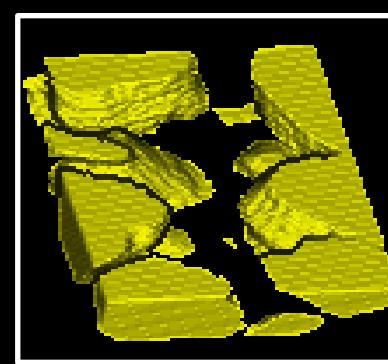
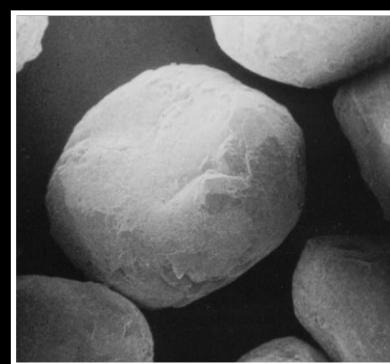
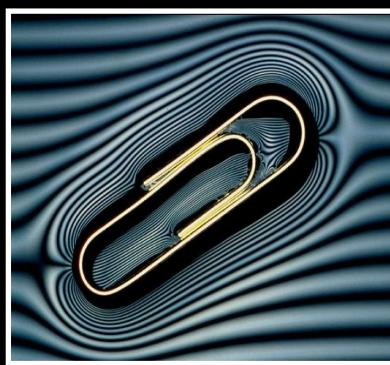
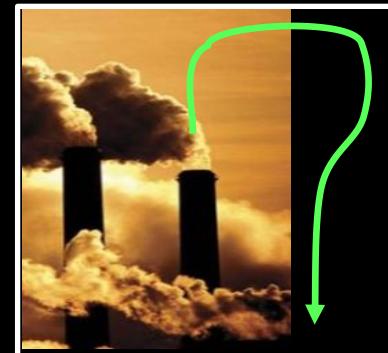
Lenses: ice, hydrate

Dissolution: Shear fractures in contraction, wormholes

May hinder long-term CO₂ geological storage

Can be used to enhance recovery

In all cases: CAUTION



Closing Thoughts

Energy: critical to life

Energy geotechnology: fascinating BTHCEM coupled problems

Unsaturated soil mechanics: great framework but careful “extension”

Methane hydrates:

C-fuel + climate + hazard

Challenging: multi-physics, testing... production

CO₂ geological storage = C-economy + climate change

Must be reliable in the long time scales

Complex geo-plumbing

Faustian bargain ?

Emergent phenomena & unanticipated coupled processes

Caution: positive feedback in coupled processes

Various localizations can be anticipated



F. Franciscas



T.S. Yun



J.Y. Lee



A.I. Martin



D. Cortes



T.H. Kwon



P. Taboada



C.H. Lee



J.W. Jung



H. Shin



N. Espinoza



S. Dai



C. Ruppel
USGS



C Tsouris
ORNL and team



G.C. Cho
KAIST



M. Sanchez
Texas A&M



K. Soga
Cambridge U.



JS Lee
Korea U.